

# Three sources of CO<sub>2</sub> efflux from soil partitioned by <sup>13</sup>C natural abundance in an incubation study<sup>†</sup>

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This study describes a novel approach to separate three soil carbon (C) sources by one tracer method (here <sup>13</sup>C natural abundance). The approach is based on the combination of C<sub>3</sub> and C<sub>4</sub> sources in different treatments, identical decomposition of C<sub>3</sub> and C<sub>4</sub> substances in soil, and subsequent calculation of their contribution to the total CO<sub>2</sub> efflux. We used the temporal dynamics of the CO<sub>2</sub> efflux from a C<sub>3</sub> grassland soil amended with added C<sub>3</sub> or C<sub>4</sub> slurry and/or C<sub>3</sub> or C<sub>4</sub> sugar to estimate contributions of three separate C sources: native soil organic matter (SOM), slurry and sugar, to CO<sub>2</sub> efflux. Soil with slurry and/or sugar was incubated under controlled conditions, and concentration and  $\delta^{13}$ C values of evolved CO<sub>2</sub> were measured over a 2-week period. The main assumption needed for separation of three C sources in CO<sub>2</sub> efflux, i.e. identical decomposition of applied C<sub>3</sub> and  $C_4$  sugars in soil, was investigated and proven. The relative contribution to the CO<sub>2</sub> efflux increased, but its duration decreased with an increased microbial availability of the C source, i.e. sugar > slurry > SOM. The microorganisms used the C sources according to their availability. The contribution of sugar to the CO<sub>2</sub> efflux was finished after 2-4 days. Separation of three CO<sub>2</sub> sources and comparison of CO<sub>2</sub> from different treatments tracing the changes of SOM and slurry decomposition induced by addition of sugar were investigated. During the sugar decomposition (the first 2–4 days), the SOM decomposition strongly decreased. At the same time the contribution of slurry-C to CO<sub>2</sub> increased. The shortcomings and limitations as well as possible future applications of the suggested method including FACE (Free Air Carbon dioxide Enrichments) and continuous labelling experiments are discussed. Copyright © 2005 John Wiley & Sons, Ltd.

The <sup>13</sup>C natural abundance method has frequently been used in the last two decades to elucidate the distinct C<sub>3</sub> and C<sub>4</sub> plant-derived sources of C in soil organic matter (SOM). The principles, advantages, limitations and shortcomings of the <sup>13</sup>C natural abundance have been described previously.<sup>1–3</sup> The method has quantified the flows of plant- or slurryderived C in SOM studies of the bulk soil,<sup>4</sup> particle size fractions,<sup>5,6</sup> rhizosphere,<sup>7,8</sup> microbial biomass and water-soluble carbon,<sup>9,10</sup> and soil CO<sub>2</sub> efflux.<sup>11,12</sup>

A limitation of <sup>13</sup>C natural abundance has been the same as with any other single tracer method (e.g. artificial <sup>14</sup>C or <sup>13</sup>C labelling), that only two C sources can be separated in one C pool (e.g. SOM) or C flow (e.g. CO<sub>2</sub>). However, the complexity and interdependence of soil C flows and food webs urgently requires concurrent quantification of more than two C sources. In view of this we have endeavoured to develop an approach to separate three C sources by one tracer method (here using <sup>13</sup>C natural abundance). We used the CO<sub>2</sub> efflux

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evolved from soil by individual and combined incubation with slurry and sugar. The main hypothesis was that multiple combinations of three C sources (soil, slurry, and sugar) with different isotopic signature (originated from C<sub>3</sub> or C<sub>4</sub> vegetation) would allow an estimation of the three separate C contributions to the CO<sub>2</sub> efflux. In this study we describe our conceptual approach on how the  $\delta^{13}$ C natural abundance tracer technique could distinguish and quantify the short-term CO<sub>2</sub> release from three C sources: SOM, slurry and sugar.

#### **EXPERIMENTAL**

#### Soil sampling and preparation

Soil (Dystric Gleysol, clayey non-calcareous) material was collected from a permanent grassland pasture (>40 years), Rowden Moor, located near North Wyke, Devon, southwest England (50°45′ N, 4°53′ W). The dominant vegetation was *Lolium perenne* L. with varying amounts of *Cynosurus, Festuca, Agrostis, Holcus* and *Dactylis* spp. Application of mineral fertilizers has not occurred at the site since 1995, but cattle and sheep have grazed there. A detailed description of the soil was presented earlier.<sup>12,13</sup> Three subsites were randomly

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sampled and isolated intact soil cores  $(20 \times 25 \text{ cm}^2)$  from the topsoil (0–10 cm) were stored at 5°C. The upper 0–4 cm contained too many roots and was not used in the experiment. The remaining soil (4–10 cm depth) was then field moist sieved (<7 mm) to remove bigger roots and stones. We originally weighed 118 g of the soil with a C content of 5.09% into Kilner jars, corresponding to 6g of soil C ( $\delta^{13}$ C = –28.6‰), to which we added 2 g slurry-C. No slurry, C<sub>3</sub> slurry ( $\delta^{13}$ C = –30.7‰) or C<sub>4</sub> slurry ( $\delta^{13}$ C = –21.3‰) was applied to the soil on 26 January 2001. The samples had a soil moisture of 70% field capacity (FC) and were incubated at 27°C for 40 days. The results of the CO<sub>2</sub> fluxes and their partitioning into CO<sub>2</sub> derived by decomposition of SOM or slurry have been published<sup>12</sup> and will not be described here. After this incubation, the soil was dried and left for 2 years.

#### Experiment design and soil incubation

We established a two by three factorial experiment. The first factor was slurry application. Soil + No slurry, Soil + C<sub>3</sub> slurry or Soil + C<sub>4</sub> slurry was applied to the soil on 26 January 2001.<sup>12</sup> The  $\delta^{13}$ C values of these samples were measured again prior to the start of this specific experiment and found to be -28.5, -29.0 and -26.6‰, respectively. On April 4 2003 we filled nine 150-mL jars with 15 g soil for each of these three treatments. The soil was moistened to 70% of FC and preincubated for 2 weeks. The second factor was sugar application. No sugar, C<sub>3</sub> sugar or C<sub>4</sub> sugar was applied to each soilslurry treatment. The  $\delta^{13}$ C value of the (sugar beet) C<sub>3</sub> sugar was -27.2%, and the (sugar cane) C<sub>4</sub> sugar was -11.0%. The amount of sugar-C applied corresponds to 1% of previously applied slurry-C and to 0.3% of SOM. This sugar-C (2.544 mg C=6.36 mg sugar) dissolved in 1 mL distilled water was added to 15 g soil. The control soil did not receive any slurry or sugar. So, the final experiment design included nine treatments (three replicates each), different in slurry or sugar application.

The moisture contents of the soil and both slurries were determined by weighing before and after drying in an oven at 85°C. Total C and N content of soil, slurry and sugar samples was determined using a CHN auto-analyzer (Carlo Erba NA2000, Milan, Italy). The  $\delta^{13}$ C values were analyzed at IGER using a continuous flow ANCA SL 20/20 system (Europa, Crewe, UK). Natural abundances of <sup>13</sup>C were expressed as  $\delta^{13}$ C (‰) versus VPDB. The analytical precision of all the  $\delta^{13}$ C measurements was 0.1‰.

Periodically (on days 1, 2, 3, 4, 6, 7, 10, and 11 after sugar application, Table 1) the jars were flushed with CO<sub>2</sub>-free air and sealed with a greased rubber ring and a lid with a septum for the needle. Between 2 (at the beginning) and 4 (at the end of incubation) h after closing the 150-mL jars, air samples (12 mL) were taken from the headspace using evacuated Exetainers and analyzed for their CO<sub>2</sub> concentration and  $\delta^{13}$ C value using gas chromatography/isotope-ratio mass spectrometry (GC-IRMS; Europa, Crewe, UK). The resulting CO<sub>2</sub> concentration and isotopic content were corrected for the small amount of ambient air (measured at 130 (±<20) ppm CO<sub>2</sub> and  $\delta^{13}$ C of -7.5%) remaining in the jars after flushing. We did not measure the CO<sub>2</sub> efflux continuously. Therefore, the results are presented as CO<sub>2</sub> efflux rate, and not as a cumulative CO<sub>2</sub> efflux. The ppm CO<sub>2</sub> data measured by GC



**Table 1.** Relative (percentage  $\pm$  SE) contribution of C<sub>4</sub> source to the total CO<sub>2</sub> efflux from the soil calculated by <sup>13</sup>C natural abundance before and after substrate additions

C <sub>4</sub> source	Contribution			
Slurry	C <sub>4</sub>	_	C <sub>4</sub>	C <sub>3</sub>
Sugar	_	C <sub>4</sub>	C <sub>3</sub>	C <sub>4</sub>
Sampling hours				
17	$46\pm11$	$74\pm11$	$108\pm23$	$60\pm11$
42	$46\pm14$	$46\pm33$	$108\pm23$	$45\pm33$
68	$35\pm19$	$21\pm54$	$86\pm22$	$27\pm54$
96	$42\pm9$	$18\pm44$	$92\pm17$	$23\pm44$
139	$34\pm5$	$14\pm73$	$86\pm5$	$18\pm73$
169	$34\pm17$	$15\pm95$	$81\pm29$	$17\pm95$
233	$23\pm25$	$14\pm96$	$64\pm23$	$11\pm96$
258	$24\pm28$	$18\pm129$	$67\pm23$	$9\pm129$

were recalculated to mass units according to the ideal gas law and all results of  $CO_2$  efflux are presented as  $\mu g C g^{-1}$  soil  $h^{-1}$ .

### Calculations

Nine different combinations of soil, slurry, and sugar resulted in a maximum of three C sources of respired  $CO_2$  presented in each jar: SOM (C<sub>3</sub> only), slurry-C (C<sub>3</sub> or C<sub>4</sub>), and sugar-C (C<sub>3</sub> or C<sub>4</sub>). To calculate the contribution of C<sub>4</sub> (slurry- or sugar-derived) C from the cases with **two C sources**, the following standard equation was used:

$$C_4^* = C_t \times (\delta_t - \delta_3) / (\delta_4 - \delta_3) \tag{1}$$

where  $C_t = C_3^* + C_4^*$  is the total CO<sub>2</sub> efflux from soil,  $C_3^*$  is the amount of CO<sub>2</sub>-C derived from the C<sub>3</sub> soil,  $C_4^*$  is the amount of CO<sub>2</sub>-C derived from C<sub>4</sub> slurry or sugar,  $\delta_t$  is the  $\delta^{13}$ C value of the C<sub>t</sub> of CO<sub>2</sub>,  $\delta_4$  is the  $\delta^{13}$ C of the C<sub>4</sub> slurry (=-21.3‰) or C<sub>4</sub> sugar (=-11.0‰), and  $\delta_3$  is the  $\delta^{13}$ C of the C<sub>3</sub> soil ( $\delta^{13}$ C of soil was -28.5‰).

To calculate the contribution of SOM  $(C_3)$  carbon from the cases with two C sources, the following equation was used:

$$C_{3}^{*} = C_{t} - C_{4}^{*} \tag{2}$$

In the treatments with three  $CO_2$  sources, with either slurry <u>or</u> sugar originating from  $C_4$  vegetation, the contribution of the  $C_4$  source was calculated according to Eqn. (1), and the contribution of the sum of two  $C_3$  sources was calculated according to Eqn. (2). The closeness of the  $\delta^{13}C$ values of the soil and  $C_3$  sugar suggested that the error of calculation of the contribution of  $C_4$  slurry by this method is rather small. The differences in the  $\delta^{13}C$  values of the soil and  $C_3$  slurry (2.2‰) could mean that the calculated contribution of  $C_4$  sugar can be biased. Therefore, we used the  $\delta^{13}C$  value of soil +  $C_3$  slurry mixture of -29.0‰ measured before the incubation (21.5% slurry-C + 78.5% soil-C). Possible shortcomings of this approach are discussed below.

In the treatments with **three CO**<sub>2</sub> **sources**, **with slurry and sugar both originating from C**<sub>4</sub> **vegetation**, the contribution of the sum of both C<sub>4</sub> sources to total CO<sub>2</sub> efflux could not be calculated according to Eqn. (1), because C<sub>4</sub> sugar and C<sub>4</sub> slurry have different  $\delta^{13}$ C values. We used an indirect calculation based on the following assumptions:



- There were no differences between decomposition of C<sub>3</sub> and C<sub>4</sub> sugars and therefore in the contribution of C<sub>3</sub> and C<sub>4</sub> sugars to the CO<sub>2</sub> efflux,
- 2. C<sub>3</sub> and C<sub>4</sub> sugars have the same effect on SOM decomposition,
- 3.  $C_3$  and  $C_4$  sugars have the same effect on slurry decomposition,
- 4. The slurry type  $(C_3 \text{ or } C_4)$  has the same effect (or no effect) on sugar decomposition.

The assumptions 1, 2, and 3 are obvious, because chemically the origin of the sugar (very small differences in isotopic composition) should not affect biochemical transformations. Nevertheless, these assumptions were tested with respect to the  $CO_2$  efflux (see below). The last assumption concerning the same effect of two slurry types on sugar decomposition could not be fully tested within the current study.

Based on these assumptions, the contribution of  $C_4$  sugar ( $C_4^*$  <sub>Sugar</sub>) in the treatment with  $C_4$  slurry **and**  $C_4$  sugar addition was then taken from the treatment with  $C_3$  slurry **and**  $C_4$  sugar addition. The contribution of  $C_4$  slurry ( $C_4^*$  slurry) in the treatment with  $C_4$  slurry **and**  $C_4$  sugar addition were taken from the treatment with  $C_4$  slurry **and**  $C_3$  sugar addition. The contribution of SOM ( $C_3^*$ ) in the treatment with  $C_4$  slurry **and**  $C_3$  sugar addition. The total  $CO_2$  efflux ( $C_t$ ):

$$C_3^* = C_t - C_4^*_{Sugar} - C_4^*_{Slurry}$$
 (3)

The experiment was conducted with three replicates. The standard deviation (SD<sub>pC4</sub>) of the proportion of C<sub>4</sub> source in CO<sub>2</sub> efflux was calculated from the standard deviations of  $\delta^{13}$ C values of CO<sub>2</sub> efflux from sample soil (SD<sub>5</sub>) and from reference (SD<sub>R</sub>), as well as  $\delta^{13}$ C values of CO<sub>2</sub> efflux of sample ( $\delta_{S}$ ) and reference ( $\delta_{R}$ ):<sup>6</sup>

$$SD_{pC_4} = \sqrt{\left(\frac{SD_S}{\delta_S - \delta_R}\right)^2 + \left(\frac{SD_R}{\delta_S - \delta_R}\right)^2}$$
 (4)

The standard deviation (SD) of values obtained as differences between total  $CO_2$  efflux and  $C_4$ -derived  $CO_2$  was calculated by using the SD of both parameters:

$$SD = \sqrt{(SD_{CO_2})^2 + (SD_{C_4})^2}$$
 (5)

where  $SD_{CO_2}$  and  $SD_{C_4}$  are standard deviations of total and C<sub>4</sub>-derived CO<sub>2</sub> efflux, respectively. For the calculation of the SD of C<sub>3</sub><sup>\*</sup> values obtained by difference according to Eqn. (3), the following equation was used:

$$SD = \sqrt{(SD_{CO_2})^2 + (SD_{C_4 \text{ Sugar}})^2 + (SD_{C_4 \text{ Slurry}})^2}$$
 (6)

Standard errors are presented in the figures.

#### RESULTS

## The CO<sub>2</sub> efflux from the soil and $\delta^{13}$ C of CO<sub>2</sub>

Prior to the sugar application (t = 0), the soil  $CO_2$  efflux was in the  $C_4$  slurry treatment ca. 2 and 1.5 times higher than in the unamended and the  $C_3$  slurry treatment, respectively (Fig. 1, top). Different  $CO_2$  efflux rates of  $C_3$  and  $C_4$  slurries were

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**Figure 1.** Total CO<sub>2</sub> efflux (±SE) from the soil amended with C<sub>4</sub> slurry (top) or C<sub>4</sub> sugar (bottom) and the contribution of SOM and one of two different C<sub>4</sub> sources (slurry or sugar) to the CO<sub>2</sub> efflux calculated directly by its  $\delta^{13}$ C value.  $\blacksquare$  or  $\bullet$  total CO<sub>2</sub> efflux;  $\triangle$  CO<sub>2</sub> from C<sub>3</sub> source (=SOM);  $\Diamond$  total CO<sub>2</sub> efflux from the soil without any addition of C<sub>4</sub> source (=control);  $\thickapprox$  total CO<sub>2</sub> (2): the total CO<sub>2</sub> efflux from the soil amended with C<sub>3</sub> sugar is shown.

known to occur beyond 10 days after slurry addition<sup>12</sup> and this showed that the  $C_3$  and  $C_4$  slurries were not identical. Therefore, the  $C_3$  slurry contribution was not equal to the calculated contribution of the  $C_4$  slurry to the total  $CO_2$ efflux. So, it was not possible to calculate directly the relative contribution of  $C_3$  slurry to the total  $CO_2$  efflux based on the values of the contribution from the  $C_4$  slurry as was done for  $C_3$  and  $C_4$  sugars (see below).

The CO<sub>2</sub> efflux strongly increased in all treatments on day 1 (17 h) after sugar addition (Fig. 1, bottom). The CO<sub>2</sub> efflux then rapidly decreased, but remained at nearly double the level of the flux before the sugar addition between days 2 and 4. Beyond 7 days after sugar addition, the total  $CO_2$  efflux from soil without slurry was no different from that before sugar addition (Fig. 1, bottom). One important assumption is necessary to distinguish between the three C sources in the  $CO_2$  efflux, i.e. that there were no significant differences in decomposition of C3 and C4 sugars. Using the chosen experimental approach, we can compare the decomposition of C<sub>3</sub> and C<sub>4</sub> sugars only based on the total CO<sub>2</sub> efflux from soils. Indeed, the curves of total CO<sub>2</sub> efflux from soils treated with  $C_3$  or  $C_4$  sugars do not differ significantly (p > 0.05) at any time, and except on day 2 the curves were nearly identical (Fig. 1, bottom). We therefore concluded that the C decomposition of the sugar beet and sugar cane sugars was similar and that the C3 sugar contribution was equal to the



**Figure 2.** Dynamics of  $\delta^{13}$ C values (‰, ±SE) of CO<sub>2</sub> efflux after addition of C<sub>4</sub> sugar to soil previously amended with slurry:  $\bullet$  no slurry;  $\blacktriangle$  C<sub>3</sub> slurry;  $\blacksquare$  C<sub>4</sub> slurry. The level of X scale shows the  $\delta^{13}$ C value of C<sub>4</sub> sugar (11.0‰).

# calculated contribution of the $C_4$ sugar to the total $CO_2$ efflux.

The  $\delta^{13}$ C values of CO<sub>2</sub> efflux show the contribution of C<sub>4</sub> sources (sugar or slurry). The addition of C<sub>4</sub> sugar strongly increased the  $\delta^{13}$ C value of CO<sub>2</sub> efflux (Fig. 2). The  $\delta^{13}$ C maximum coincided with the peak CO<sub>2</sub> concentration on day 1 (17 h) after sugar addition (see Fig. 1). However, the  $\delta^{13}$ C value of the CO<sub>2</sub> then did not reach that of the C<sub>4</sub> sugar itself (-11.0%), thus indicating that the contribution of C<sub>3</sub> sources to the CO<sub>2</sub> efflux was still significant. The maximum increase in the  $\delta^{13}$ C value of CO<sub>2</sub> efflux was observed for soil unamended with slurry and the minimum for the soil amended with C4 slurry. At the same time the rate of decrease of the  $\delta^{13}$ C value of CO<sub>2</sub> efflux after the maximum showed the reverse dependence: it was fastest for soil unamended with slurry and slowest for the soil amended with C<sub>4</sub> slurry. This clearly indicates fastest utilization of easily available C source (sugar) in the soil having the smallest total amount of Corg.

The  $\delta^{13}$ C peak decreased at a slower rate than the total CO<sub>2</sub> efflux (cf. Figs. 1 and 2), indicating that some added C<sub>4</sub> sugar was not immediately decomposed to CO<sub>2</sub>, but temporarily incorporated into microbial biomass. Furthermore, it indicated that several days were necessary to replace the sugarderived C<sub>4</sub> in microbial biomass with that of C<sub>3</sub> from the SOM or slurry.

# Contribution of two C sources to total soil $CO_2$ efflux

The total CO<sub>2</sub> efflux of the two treatments C<sub>4</sub> slurry + no C<sub>4</sub> sugar and C<sub>4</sub> sugar + no C<sub>4</sub> slurry is presented in Fig. 1. The previous addition of C<sub>4</sub> slurry had nearly doubled the soil CO<sub>2</sub>, and its contribution remained significant for the first 9 days of incubation (Fig. 1, top). The C<sub>4</sub> sugar induced a peak in the total CO<sub>2</sub> efflux on days 1 and 2, and the contribution of sugar-C to total CO<sub>2</sub> was significant during the first 4 days. The contribution of C<sub>4</sub> slurry (treatment without sugar) estimated by Eqn. (1) amounted to 23–46% of the total CO<sub>2</sub> efflux (Table 1). It decreased strongly during the incubation period suggesting that slurry decomposition proceeded faster than the decomposition of the native SOM (Fig. 1, top). The estimated contribution of applied sugar (treatment without slurry) to the total CO<sub>2</sub> efflux peaked after 17 h at 74%, but it was effectively finished after 3–4 days (Table 1; Fig. 1, bottom).



**Figure 3.** Absolute contribution ( $\pm$ SE) of C<sub>4</sub> sugar to the total CO<sub>2</sub> efflux from the soil previously amended by C<sub>3</sub> slurry calculated directly by  $\delta^{13}$ C values of CO<sub>2</sub> and changes in (SOM + slurry)-C mineralization induced by sugar addition (Primed SOM + slurry C). Vertical arrows show the CO<sub>2</sub> efflux originating from C<sub>4</sub> sugar and induced priming effect. **I** total CO<sub>2</sub> efflux;  $\Diamond$  CO<sub>2</sub> efflux originated from decomposition of SOM + C<sub>3</sub> slurry from the soil after sugar addition;  $\bigcirc$  total CO<sub>2</sub> efflux from the soil amended with C<sub>3</sub> without sugar.

The direct estimation of the relative contribution of the  $C_4$  source was more difficult for three  $CO_2$  sources (Table 1, Fig. 3). For example, when  $C_3$  sugar was added to  $C_4$  slurry-amended soil, the contribution of  $C_4$  slurry was between 64–108% (Table 1). The contribution of  $C_4$  slurry to the total  $CO_2$  efflux after sugar addition was much higher than without sugar addition. It implies that sugar addition induced faster slurry decomposition (see below). The addition of  $C_4$  sugar to the  $C_3$  slurry-amended soil was not different from that of the  $C_4$  sugar application to the unamended soil (Table 1). This observation partly confirmed our fourth assumption, that the effect of slurry type ( $C_3$  or  $C_4$ ) on sugar decomposition was negligible.

The total CO<sub>2</sub> efflux evolved from the soil amended with C<sub>3</sub> slurry + C<sub>4</sub> sugar treatments is presented in Fig. 3. The soil + C<sub>3</sub> slurry (no sugar) was used as the control, in order to estimate the effect of added sugar on soil- and slurry-derived CO<sub>2</sub>.

The addition of  $C_4$  sugar to  $C_3$  slurry-amended soil strongly increased total  $CO_2$  efflux through sugar-derived  $CO_2$  (Fig. 3). These effects were similar to the sugar treatment without slurry addition (Fig. 1, bottom). Hence, the total soil  $CO_2$  efflux measured on day 1 was mainly (60%) derived from sugar-C. Note that the maximum variations and therefore uncertainties of calculation of the contribution of the  $C_4$  source were observed (1) at first sampling period when the contribution of sugar ( $C_4$ ) derived C dominated the total  $CO_2$  efflux (Figs. 1 and 3), and (2) at the last two samplings, because the contribution of sugar was negligible at this time (Table 1).

# Estimation of contribution of three C sources to the CO<sub>2</sub> efflux

For estimation of the contributions of three C sources to the total  $CO_2$  efflux from soil, two treatments were evaluated: (i) treatment with  $C_3$  slurry and  $C_3$  sugar, and (ii) treatment with  $C_4$  slurry and  $C_4$  sugar. The principles of the calculations



**Figure 4.** Principles and calculation steps for contributions of three C sources to the  $CO_2$  efflux from soil amended with  $C_3$  slurry and  $C_3$  sugar (top) and amended with  $C_4$  slurry and  $C_4$  sugar (bottom). See text for additional explanations. The calculations were done for each  $CO_2$  sampling time separately.

of contributions in both treatments were different and are presented schematically in Fig. 4.

#### Treatments with $C_3$ slurry and $C_3$ sugar

The first step in the treatment with  $C_3$  slurry and  $C_3$  sugar addition was the use of earlier calculated contribution of  $C_4$ sugar instead of that of  $C_3$  sugar (Fig. 4, top; Fig. 5, top). The second step was the estimation of the summed absolute contribution of SOM-C and slurry-C as the difference between the total  $CO_2$  efflux and the contribution of  $C_3$  sugar (Fig. 4, top). Most of the  $CO_2$  evolved on day 1 comes from  $C_3$ sugar (Fig. 5, top). The sugar contribution then decreased strongly and after day 4 it was about zero. Theoretically, if  $C_3$  and  $C_4$  slurries had identical decomposition, then the decomposition of  $C_4$  slurry +  $C_3$  sugar treatment could have been transposed into the  $C_3$  slurry +  $C_3$  sugar treatment to separate three  $CO_2$  sources in the treatment with  $C_3$  slurry and  $C_3$  sugar addition.

#### Treatments with $C_4$ slurry and $C_4$ sugar

The estimation of CO<sub>2</sub> efflux from the soil treated with C<sub>4</sub> slurry and C<sub>4</sub> sugar is more complicated and involves three steps (Fig. 4, bottom; Fig. 5, bottom). First, we separated the slurry-derived CO<sub>2</sub> by using values from the treatment with C<sub>4</sub> slurry + C<sub>3</sub> sugar addition (based upon the equal decomposition of C<sub>4</sub> and C<sub>3</sub> sugars; see above). Secondly, the contribution of C<sub>4</sub> sugar was taken from the treatment with C<sub>3</sub> slurry + C<sub>4</sub> sugar. Thirdly, the difference between the total CO<sub>2</sub> efflux and the estimated summed contributions of C<sub>4</sub> slurry + C<sub>4</sub> sugar allows the estimation of SOM-derived CO<sub>2</sub> (Fig. 4, bottom).

Most of the  $CO_2$  emitted during the first 3 days after the sugar addition was derived from the sugar; however, the slurry-C contribution was also important (Fig. 5, bottom). The contribution of SOM was marginal during the first 7 days. However, because of the uncertainty of the  $CO_2$  sources



**Figure 5.** Absolute contribution  $(\pm SE)$  of three C sources to the total CO<sub>2</sub> efflux from the soil calculated indirectly, considering identical decomposition of C<sub>3</sub> and C<sub>4</sub> sugars. The treatments of slurry and sugar are shown in the box in the center of each graph. Vertical arrows show the C<sub>3</sub> or C<sub>4</sub> source. If total CO<sub>2</sub> efflux;  $\triangle$  CO<sub>2</sub> from SOM (C<sub>3</sub> source);  $\neq$  CO<sub>2</sub> from the sum of SOM and slurry;  $\Diamond$  total CO<sub>2</sub> efflux from the respective treatment with addition of C<sub>4</sub> source;  $\blacklozenge$ soil without any additions.

separation during the first 2 days, we cannot exactly estimate the contribution of sugar- and slurry-derived C.

### DISCUSSION

#### Partitioning of two and three C sources

The two C source partitioning of the total CO<sub>2</sub> efflux (Table 1, Fig. 1) was calculated in the same way as in many previous studies.<sup>1,2,7,8,12,17,18</sup> It showed agreement with other results that the contribution to the CO<sub>2</sub> efflux increased, but the duration of the contribution decreased with an increased microbial availability of the C source, i.e. SOM < slurslurry < sugar.<sup>10,12,19–21</sup> To our knowledge, this is the first time that the contribution of three C sources (SOM, slurry and sugar) of CO<sub>2</sub> efflux has been separated by using only one tracer method (Fig. 5). The separation of three C sources was achieved with a simple experimental design: by comparing the temporal trends in the concentration and  $\delta^{13}$ C value of the CO<sub>2</sub> efflux from one soil with three 'past' treatments (no, C<sub>3</sub> and C<sub>4</sub> slurry) and three 'newly' imposed treatments (no,  $C_3$  and  $C_4$  sugar). Based on the existing general assumption regarding similar decomposition of C<sub>3</sub> and C<sub>4</sub> substrates in soil,<sup>1</sup> we successfully separated the contribution of the three C sources to the CO<sub>2</sub> efflux. We did add some additional assumptions, as the  $CO_2$  efflux from the  $C_3$  and  $C_4$  slurry was found to be different. As the decomposition of C3 and

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 $C_4$  sugars was indeed identical, the contribution of  $C_4$  sugar to the total  $CO_2$  efflux from soil amended with or without  $C_3$ slurry is equal to the contribution of  $C_3$  sugar. The three C source separation using one C tracer was achieved by experimental design and not through the tracer method itself. Furthermore, either <sup>13</sup>C or <sup>14</sup>C can be used as the single C tracer. However, if <sup>13</sup>C and <sup>14</sup>C are combined, the necessity of complicated experimental design can be omitted and no assumptions are necessary.<sup>7,8</sup> Furthermore, there is the tantalizing prospect of using our single tracer three C source design and a combination of <sup>13</sup>C and <sup>14</sup>C tracers to separate four or more C sources simultaneously. This could open new ways for C studies in soils and other complex systems.

# Changes in SOM and slurry decomposition after addition of sugar

The application of organic substances, which are more easily available for microorganisms than SOM (e.g. slurry or sugar), change the soil microbial activity and may induce subsequent associated changes in SOM decomposition. In our experimental design, these two easily available substances (sugar and slurry) were added to soils at different times. The slurry was added to the soil much earlier than the sugar; hence, the main priming effects of SOM induced by slurry would be expected to be (nearly) finished before the application of sugar.<sup>12</sup> Addition of sugar to the soil led in most cases to a short-term (~2 days) decrease in SOM decomposition (e.g. Fig. 1, bottom; Fig. 3). Such a decrease in SOM decomposition is connected with the preferential microbial utilization of an easily available substrate such as sugar, compared with SOM.<sup>14,15</sup> It means that during the first 2 days soil microorganisms switch from the decomposition of recalcitrant SOM to the easily available sugar. Therefore, the decomposition of SOM during the first 2 days was lower than in the sugar unamended control. Such a fast switch to the easily available C source is mostly typical for microorganisms having *r*-strategy.<sup>16</sup>

The use of soil pre-treated with C4 slurry gives us a unique possibility to differentiate between three different C pools contributing to the total CO<sub>2</sub> efflux, and therefore to evaluate sources of C contributing to priming (Fig. 5, bottom). On day 1 after sugar addition, about 40% of the CO2 efflux originated from sugar and about 60% originated from slurry-C. The contribution of SOM-derived C to the total CO<sub>2</sub> efflux was marginal during the first 4-5 days. Thereafter, the contribution of SOM-derived C increased gradually and amounted to around 30% on day 11 after sugar addition. These source changes of the total CO<sub>2</sub> efflux after sugar addition can be explained by the following chain of reactions: (1) After sugar addition, the microorganisms mainly switched from the decomposition of heavily available SOM and slurry-derived C to that of sugar. During sugar utilization, microbial biomass activity and probably its in situ amount strongly increased. This rapid increase is only possible for microorganisms that have r-strategy. (2) After the sugar-C was utilized, the new microorganisms (mainly r-strategists) started to utilize the next easily available C source, which is slurry-C. Therefore, the contribution of slurry-derived C strongly dominated after day 3, and the contribution of SOMderived C was negligible during this period. (3) After about 1



week, the contribution of slurry-derived C steadily decreased and that of SOM-derived C steadily increased, nearly reaching the contribution of SOM-derived C in the treatment without sugar addition at the end of incubation (cf. Fig. 1, top). Such a succession of contributions of different C sources is a clear indicator for the preferential utilization of easily available C not only during decomposition, but also during priming.

### Difficulties and limitations of the method

The first limitation of the method is its accuracy. An important part of the separation of C sources is based on the differences between CO<sub>2</sub> efflux rates from different treatments. This difference led to the summation of the variance and resulted in higher errors than in the initial data. Such increase in the variance is especially important for SOM-derived CO<sub>2</sub> efflux calculated by the difference of three variables. So, the calculated SOM-derived  $CO_2$  efflux in the  $C_4$  slurry +  $C_4$ sugar treatment was negative at the first sampling (Fig. 5, bottom). This negative value is the result of estimation of a small value (here SOM-derived CO<sub>2</sub>) by its calculation as the difference between two large values (total CO2 and sugar-derived CO<sub>2</sub>) having relatively large errors. Therefore, the SOM-derived CO<sub>2</sub> is not significantly different from zero shortly after sugar addition. In subsequent studies, the use of cumulative systems to trap CO<sub>2</sub> (i.e. sorption on alkali, molecular sieves<sup>22,23</sup> or other techniques to measure soil CO<sub>2</sub> efflux (i.e. at constant CO<sub>2</sub> concentration<sup>24</sup>) will allow a decrease in the variation between replicates and associated error of values estimated as differences.

Another point was the background variation of  $\delta^{13}$ C values. The variation of  $\delta^{13}$ C values in SOM between the replicates is frequently higher than 0.5‰ and that of CO<sub>2</sub> efflux is about 1–1.5‰. Such variations result in errors of estimation of source contribution to the CO<sub>2</sub> efflux of around 10–15%. By calculation of some values as the difference between two others, this error will be cumulative (see above).

A separate issue is the choice of substrates with appropriate decomposition rates. The substrates we used (partly decomposed slurry and sugar) had decomposition rates that differed by more than one order of magnitude. The rapid decomposition of sugar led to a very strong increase in  $CO_2$  efflux during the first 2 days, and induced inaccuracy of estimation of SOM-derived  $CO_2$  efflux. Combination of other substances could improve the accuracy of the method.

We assumed that the isotopic discrimination by CO<sub>2</sub> production from different sources is negligible. The literature was not conclusive on this point. The  $\delta^{13}$ C of CO<sub>2</sub> efflux evolved by microbial respiration corresponds roughly to  $\delta^{13}$ C of microbial biomass.<sup>25</sup> However, by measuring  $\delta^{13}$ C of CO<sub>2</sub> respired from 21 Australian soils with C<sub>3</sub> and C<sub>4</sub> vegetation, it was shown that the microbially respired CO<sub>2</sub> is depleted on average by 2.2‰ compared with microbial biomass.<sup>26</sup> At the same time, microbial biomass was enriched by –2.0‰ compared with the  $\delta^{13}$ C of SOM. Thus, the observed <sup>13</sup>C enrichment in microbial biomass is balanced by a corresponding <sup>13</sup>C depletion in respired CO<sub>2</sub> resulting in the  $\delta^{13}$ C of respired CO<sub>2</sub> being similar to the  $\delta^{13}$ C of SOC.<sup>26</sup>

A further shortcoming affecting the accuracy is the correspondence of  $\delta^{13}C$  values between two  $C_3$  (or  $C_4)$ 



sources to calculate the contribution of the third C4 source (or C<sub>3</sub> source) by Eqn. (1). In an ideal case, the  $\delta^{13}$ C values of both C<sub>3</sub> sources should be the same. In our study, despite the close correspondence between the  $\delta^{13}$ C values of soil (-28.5%) and of C<sub>3</sub> sugar (-27.2‰), the  $\delta^{13}$ C of C<sub>3</sub> slurry (-30.7‰) was different from that of the soil. We used therefore the  $\delta^{13}$ C value of both C<sub>3</sub> sources, i.e. -29.0% (78.5% SOM +21.5%slurry), measured before the experiment, to estimate the  $C_4$ sugar contribution in C3 slurry-amended soil. The application of such a weighted  $\delta^{13}$ C value assumes an equal decomposition rate of both C3 sources. Because of the last assumption, the calculated contribution of C<sub>4</sub> sugar to the C<sub>3</sub> slurry treatment could be biased. Clearly, most shortcomings of the approach are associated with the detection level of different C sources by the <sup>13</sup>C natural abundance method. The use of <sup>14</sup>C- and/or <sup>13</sup>C-labelled substances would overcome these problems.

### **CONCLUSIONS AND FUTURE APPLICATIONS**

The combination of C<sub>3</sub> and C<sub>4</sub> source treatments (slurry and sugar) applied to a C<sub>3</sub> grassland soil allowed us to distinguish between three CO<sub>2</sub> sources, as well as the changes of C sources contributing to priming. This possibility is extremely useful for the investigation of complex natural systems (e.g. soil), which generally contain more than two C sources. However, methods reducing the initial variance of CO<sub>2</sub> efflux as well as its  $\delta^{13}$ C values are necessary to increase the significance of the observed separation of the CO2 sources. The relative contribution to the CO2 efflux increased, but the period of its contribution decreased with an increased availability of the C source, i.e. sugar  $\gg$  slurry > SOM.

The separation of three C sources in the CO<sub>2</sub> efflux by one tracer method showed advantages over the separation of two sources. Furthermore, if the C<sub>3</sub> and C<sub>4</sub> slurries had been more identical, more results would have been obtained with the same experimental effort. At this point, it makes sense to consider future possibilities, other than C<sub>3</sub> and C<sub>4</sub> sugars, of substrate or compounds with identical behaviors in soil. The first possibility is to use uniformly labelled (<sup>13</sup>C or <sup>14</sup>C) substances as analogue for an unlabelled one. The labelling of the substances does not have to be very high: a few hundred delta units would be enough to obtain significant differences. However, nearly all commercially available substances are pure chemicals. In soil and environmental studies, we mostly deal with complex substances, e.g. like plant residues, microbial biomass, etc., which are difficult to completely label in a uniform way. Another easier option would be to obtain uniformly labelled plant residues of Free Air Carbon dioxide Enrichments (FACE) experiments. These experiments<sup>27–29</sup> use CO<sub>2</sub> depleted in <sup>13</sup>C (-30% to -50%) compared with atmospheric value ( $\delta^{13}C \approx -7.5\%$ ). After

the mixing of the 'artificial'  $CO_2$  with atmosphere  $CO_2$ , the  $\delta^{13}$ C of the mixed supplied CO<sub>2</sub> is about -20% to -30%. Therefore, C<sub>3</sub> plants produced under FACE will have  $\delta^{13}$ C values of around -40% to -50%. These <sup>13</sup>C-depleted C<sub>3</sub> plant residues can be used as analogues for the same C<sub>3</sub> plants produced under normal air conditions.

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