

## Effects of nitrate and sulfate on greenhouse gas emission potentials from microform-derived peats of a boreal peatland: A $^{13}\text{C}$ tracer study



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### ABSTRACT

Increasing natural and anthropogenic deposition of nitrate ( $\text{NO}_3^-$ ) and sulfate ( $\text{SO}_4^{2-}$ ) to peatlands may modify  $\text{CH}_4$  oxidation,  $\text{CO}_2$  and  $\text{N}_2\text{O}$  production, thereby affecting the balance of greenhouse gases (GHG) globally. Among environmental factors controlling these biogeochemical processes, effects of peatland microrelief are poorly understood. Fluxes of  $\text{CO}_2$ ,  $\text{CH}_4$  and  $\text{N}_2\text{O}$  were measured before and after incubation with  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  for peat samples collected from various microrelief positions of a boreal oligotrophic mire in Eastern Finland. Soil was spiked with  $^{13}\text{C}$  to understand the processes of  $\text{CH}_4$  oxidation, its microbial utilization and incorporation into soil organic matter (SOM). We hypothesized that the addition of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  would 1) stimulate  $\text{CO}_2$  and  $\text{N}_2\text{O}$  production (nutritional effect), but 2) decrease  $\text{CH}_4$  oxidation due to acceleration of other more energetically favorable processes (e.g. denitrification), and 3) these patterns should follow the naturally established aerobic zone of a microform type and decrease with depth.

Microbial biomass (MB) at 50 cm below all microforms was 9–15 folds higher than in the topsoil. MB controlled the GHG dynamics and was related to specific depth-dependent environmental conditions, rather than oxygen availability. Indeed, production of  $\text{CO}_2$  and  $\text{N}_2\text{O}$ , and oxidation potentials of  $\text{CH}_4$  revealed no clear linkage with the naturally established aeration zone of the peatland's microforms. Following  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  addition, production of  $\text{CO}_2$  decreased by 20–65% compared to the control, with the greatest reduction in  $\text{CO}_2$  emission occurring in the topsoil of hollows. In turn,  $\text{CH}_4$  oxidation was suppressed by 20–94% with  $\text{NO}_3^-$  addition at 50 cm in lawns and with both  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  at 50 cm in hollows. The  $\text{N}_2\text{O}$  production was increased up to 180–240 times under  $\text{NO}_3^-$  treatment at 50 cm in hollows and lawns. In conclusion, human-induced deposition of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  may suppress  $\text{CO}_2$  emissions from and  $\text{CH}_4$  oxidation by boreal oligotrophic mires especially under the conditions of deposition increase. Finally, the deposition of inorganic compounds is strongly important to be considered in the estimation of ecosystem C and N balances.

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### 1. Introduction

Peatlands are a subgroup of wetlands, defined as areas with naturally accumulated peat (organic soil layer) due to a decomposition rate of organic remnants that is lower than the net primary

production (Saarnio et al., 1997). Even though peatlands cover only 4% of the land surface (some 4 million  $\text{km}^2$ ) they contain around 20% of the global terrestrial carbon (C) stocks (Gorham, 1991; Roulet, 2000). Peatlands can be considered as  $\text{CO}_2$  sinks due to the sequestration of atmospheric  $\text{CO}_2$ , but on the other hand, they are  $\text{CH}_4$  sources due to the process of methanogenesis and  $\text{CH}_4$  emissions (Gorham, 1991; Lafleur et al., 1997; Saarnio et al., 1997; Ye et al., 2012). Since  $\text{CO}_2$  and  $\text{CH}_4$  are important greenhouse gases (GHG) that contribute to global warming (IPCC, 2014), and the northern peatlands cover the area of ca. 3.7 million  $\text{km}^2$  in total (Yu, 2012), understanding  $\text{CO}_2$  production and  $\text{CH}_4$  oxidation processes in boreal peatlands is essential for estimating the global C budget.

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Nitrous oxide (N<sub>2</sub>O) is another important GHG, which is emitted in quantities up to three orders of magnitude smaller than CO<sub>2</sub> emissions, in absolute terms. Though N<sub>2</sub>O warming influence (in W m<sup>-2</sup>) was ca. 6.4% in 2015 from the total GHG (NOAA, 2015), its cumulative forcing of the global warming potential over 100-year time frame is 265 per kg pulse as compared to CO<sub>2</sub> (global warming potential 1) (IPCC, 2014). N<sub>2</sub>O now has the third largest forcing of the anthropogenic gases, at 0.17 ± 0.03 W m<sup>-2</sup> an increase of 6% since 2005 (Myhre et al., 2013). Moreover, N<sub>2</sub>O is the most important gas in terms of stratospheric ozone destruction (Regina et al., 1996; Marushchak et al., 2011).

Generally, CO<sub>2</sub> and N<sub>2</sub>O production and CH<sub>4</sub> oxidation in peatlands are controlled by a number of environmental parameters, such as water table (WT) level, temperature and plant communities (Lai, 2009). Among other factors, deposition of some anions, such as sulfate (SO<sub>4</sub><sup>2-</sup>) and nitrate (NO<sub>3</sub><sup>-</sup>), may affect GHG fluxes (Eriksson et al., 2010; Sutton-Grier et al., 2011). Supply of peatlands with SO<sub>4</sub><sup>2-</sup> occur through air pollution, intensive volcanic activity, mineral weathering and acidic deposition from the atmosphere, whereas NO<sub>3</sub><sup>-</sup> inputs originate from the anthropogenic eutrophication of inland waters and/or acidic deposition from the atmosphere (Sutton-Grier et al., 2011). SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> anions have two main functions related to the GHG balance: (i) they serve as nutrients, stimulating plant growth and rhizodeposition (Kuz'yakov and Domanski, 2000) and microbial activity (Blagodatskaya et al., 2010) and (ii) they participate in redox reactions as alternative electron acceptors (AEAs) when oxygen availability is low. The presence of AEAs can reduce CH<sub>4</sub> production (Bodegom and Stams, 1999; Eriksson et al., 2010; Smemo and Yavitt, 2011; Segarra et al., 2013). This is due to a combination of inhibition and competitive effects between organisms which use AEAs and methanogens for electron donors (Bodegom and Stams, 1999; Eriksson et al., 2010). SO<sub>4</sub><sup>2-</sup> has been reported to suppress methanogenesis (Gauci et al., 2004) or to have no effects on CH<sub>4</sub> production (Vile et al., 2003a,b). There is a lack of evidence that SO<sub>4</sub><sup>2-</sup> amendments affect aerobic CH<sub>4</sub> oxidation (Eriksson et al., 2010).

N<sub>2</sub>O is produced in soils as a facultative byproduct of aerobic nitrification and anaerobic denitrification (Goldberg et al., 2010; Marushchak et al., 2011). Waterlogged conditions are usually associated with low N<sub>2</sub>O production due to the low rates of nitrification and, subsequently, denitrification. In contrast, when water table is lowered, resulting in strong acceleration of OM mineralization leading to mineral N release, N<sub>2</sub>O emission rates also increase. (Goldberg et al., 2010).

Prevailing environmental factors in peatlands (WT level, temperature, vegetation, water chemistry) are strongly interrelated, leading to formation of specific local conditions on the peatland surface and development of microrelief forms – so-called microform types. The three microform types are: elevated hummocks, depressed hollows and intermediate lawns (Dorodnikov et al., 2011, 2013 and references therein). The water table level relative to the soil surface varies between microforms, increasing in the order hummocks < lawns < hollows. This results in variable thickness of the oxidative zone and the growth of distinct vegetation communities. The existing differences between microforms promote the formation of specific microbial populations, thereby affecting GHG fluxes (Kotiaho et al., 2013; Deng et al., 2014). The patterns of GHG dynamics also change with peat depth, mostly due to the temperature decrease, low oxygen availability and limited access to fresh plant-derived inputs (Dorodnikov et al., 2013).

In the current study, we tested the effects of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> addition on CO<sub>2</sub> and N<sub>2</sub>O production as well as potential CH<sub>4</sub> oxidation in a peat soil (Histosol), and their dependence on peatland microforms and peat depth. We modeled the situation of lowering of the WT, thus promoting aerobic soil organic matter

(SOM) decomposition and CH<sub>4</sub> oxidation. We put forward the following hypotheses: (i) microbially-driven GHG production (CO<sub>2</sub>, N<sub>2</sub>O) and oxidation (CH<sub>4</sub>) should follow the naturally established aerobic zone of a microform type and increase from hollows to lawns and further to hummocks; (ii) CO<sub>2</sub> and N<sub>2</sub>O production and CH<sub>4</sub> oxidation should decrease with depth due to *in situ* decreasing availability of oxygen and fresh rhizodeposits; (iii) addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> should stimulate CO<sub>2</sub> and N<sub>2</sub>O production (nutritional effect), but decrease CH<sub>4</sub> oxidation due to acceleration of other more energetically favorable processes (e.g. nitrification-denitrification).

## 2. Materials and methods

### 2.1. Study site and peat sampling

The experimental site was an ombrotrophic minerogenic fen Salmisuo in Eastern Finland (62°47'N, 30°56'E). The site is described in detail elsewhere (Alm et al., 1999; Becker et al., 2008; Saarnio et al., 1997). The surface of the experimental site was subdivided into three relief microform types according to the topography, water table level and vegetation communities: 1) dry and elevated hummocks with average WT around -20 cm (below peat surface) and dominant plant species *Eriophorum vaginatum*, *Pinus sylvestris*, *Andromeda polifolia*, *Sphagnum fuscum*; 2) intermediate lawns with average WT from -5 to -15 cm (dominant plants *Eriophorum vaginatum*, *Sphagnum balticum*, *Sphagnum papillosum*) and 3) the most wet - hollows with the average WT between 0 and 5 cm above the peat surface (dominant plants *Scheuchzeria palustris*, *Sphagnum balticum*) (Becker et al., 2008; Dorodnikov et al., 2013). The soil of the Salmisuo peatland could be classified as a Dystric Histosol (WRB, 2014) with an organic layer up to 2–2.3 m depth, consisting predominantly of *Sphagnum* remnants. The soil was randomly sampled from each microform type of a study site (50 × 50 m) from 3 depths (15, 50 and 200 cm) using a stainless-steel peat auger with a gouge-with-flap principle (Eijkelpoort Agrisearch Equipment, Giesbeek, Netherlands). Samples (81 altogether) were collected in plastic bags, trapped air was maximally removed and bags were transported in a thermobox to the laboratory, where they were kept tightly closed for 20 days at low temperature (4 °C) in darkness until the experiment. The natural moisture content of the samples comprised 90–95% of the peat fresh weight and generally decreased with depth. The pH varied between 3.9 and 4.6 in all microforms and increased with depth.

### 2.2. Experiment set-up

The following treatments were included in order to estimate the effect of microforms, depth, and the addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> on CO<sub>2</sub> and N<sub>2</sub>O production and CH<sub>4</sub> oxidation under aerobic conditions. These were:

- (i) Addition of NO<sub>3</sub><sup>-</sup> as KNO<sub>3</sub> and SO<sub>4</sub><sup>2-</sup> as Na<sub>2</sub>SO<sub>4</sub> in final concentrations chosen after Smemo and Yavitt (2007): NO<sub>3</sub><sup>-</sup> (10 mM) and SO<sub>4</sub><sup>2-</sup> (1 mM) (corrected for the dilution with the natural soil moisture content). A control treatment without anions (amended with ultrapure deionized water) and soil-free blank (pure water of similar volume to soil) were included to follow the process of gas sampling and physical gas leakage. All the solutions and water were added in volume of 0.5 ml, thus increased total moisture content by ca. 3% from the natural soil moisture content (on the weight basis).
- (ii) Addition of labeled <sup>13</sup>CH<sub>4</sub> (5 atom% with a headspace concentration of 0.5–0.6%) to all treatments with and without NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> amendments.

Soil samples (15–20 g of fresh weight) were put into 100-ml glass bottles with wide necks. The bottles were closed with tight-fitting butyl rubber septa and screw caps. The measurements of CO<sub>2</sub>, N<sub>2</sub>O and CH<sub>4</sub> along with <sup>13</sup>C-isotope signatures were done microform by microform. The duration of the incubation period lasted 7–10 days after addition of <sup>13</sup>CH<sub>4</sub>, NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>. During the experiment, all microcosms were stored in a dark room at a temperature of 21–23 °C. Gas fluxes (CO<sub>2</sub>, CH<sub>4</sub>) and δ<sup>13</sup>C values of CO<sub>2</sub> were measured on a cavity ring-down spectroscope (CRDS) Picarro G2131-i (Picarro, Inc. Santa Clara, CA, USA) by injection of gas headspace subsamples (1 ml + 59 ml N<sub>2</sub>) with syringes. Its operational range for CO<sub>2</sub>: 0.01–0.4%; for CH<sub>4</sub>: 0–1000 ppm; precision of δ<sup>13</sup>C in CO<sub>2</sub>: 0.1–0.25‰. For N<sub>2</sub>O analysis, microcosm headspace was sampled with a 1-ml syringe, gas samples were transferred to 12-ml evacuated and N<sub>2</sub>-flushed glass vials, diluted with 20 ml N<sub>2</sub>, and measured on a gas chromatograph GC 6000 (Carlo Erba Instruments) equipped with ECD and FID detectors.

### 2.3. Microbial biomass C and bulk soil C, N and δ<sup>13</sup>C measurements

Microbial biomass was measured based on extracted total DNA using FastDNA Spin Kit for Soil (BIO 101/Qbiogene, MP Biomedicals), according to the protocol of the manufacturer for the FastPrep-24 TM instrument (MP Biomedicals, LLC, Santa Ana, CA, USA). Briefly, soil samples (0.2 g fresh weight) were put into Lysing Matrix E tubes and 978 μl sodium phosphate buffer and 122 μl of lysis solution buffer were added. Suspended soil was homogenized in the FastPrep-24 for 40 s at a speed level 6.0. Lysing Matrix E tubes were centrifuged at 14,000×g (Centrifuge 5416, Eppendorf AG, Hamburg, Germany) for 10 min and the supernatant was transferred to other microcentrifuge tubes (2.0 ml). 250 μl of protein precipitation solution was added and mixed well by shaking. The 2-ml tubes were centrifuged again for 5 min and supernatant was transferred to 15-ml centrifuge tubes, where 1.0 ml of binding silica matrix suspension was added. Tubes were put on a shaker for 5 min to allow the DNA binding. About 500 μl of the suspension from the top was discarded and the remainder was transferred into SPIN-TM filters and centrifuged at 14,000×g for 1 min. Prepared SEWS-M wash solution (500 μl) was added to the filters. They were centrifuged a second time at 14,000×g for 2 min to dry the matrix of the residual washing solution. SPIN-TM filters were air dried for 5 min at room temperature and 150 μl DES (DNase/Pyrogen-Free water) was added. Filters were placed in a heat block for 5 min at 55 °C. Finally, SPIN-TM filters were centrifuged for 1 min and the supernatant in catch tubes was the extracted DNA. The extracts were either stored in a freezer at –20 °C or at 55 °C when measured immediately. The quantity of DNA extractions was detected with a PicoGreen (Invitrogen TM, Life Technologies GmbH, Darmstadt, Germany) and TE buffer (Tris EDTA, MP Biomedicals) solutions after necessary dilution. Measurement was carried out in 96-well black polystyrene microplates (Brand GmbH, Wertheim, Germany) on a Victor<sup>3</sup> 1420-050 Multilabel Counter (Perkin Elmer, Waltham, MA, USA) according to a protocol with fluorescence excitation at 485 nm and emission at 535 nm (1.0s). Based on measured standards, the amount of DNA was calculated and converted to microbial biomass C (μg g<sup>-1</sup> of dry soil) according to Blagodatskaya et al. (2014).

Bulk soil C and N contents were measured in dried and ground peat samples on an ElementarVario EL Cube CN analyzer (ElementarAnalysensysteme GmbH, Hanau, Germany). Values are presented as mg C (N) per gram of dry soil (mg g<sup>-1</sup> soil DW). Stable C isotope composition was measured in the same soil samples at the Competence Center for Stable Isotopes (KOSI, University of Göttingen, Germany) on an Isotope Ratio Mass Spectrometer

(IRMS) Delta V Advantage with a ConFlo III interface (Thermo Electron, Bremen, Germany) equipped with an elemental analyzer Flash 2000 (Thermo Fisher Scientific, Cambridge, UK).

### 2.4. Calculations and statistics

Gross CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes were estimated from the linear rate of change in CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O concentrations overtime. The Ideal Gas Law was used to convert the raw data from the instrument (in ppm) to mass units:

$$n = P \cdot V / R \cdot T \quad (1)$$

where n is the amount of gas (in moles), P is the atmospheric pressure (101.325 kPa), V is the volume of gas in the headspace (L), R is the Ideal Gas Constant (8.31 J K<sup>-1</sup>mol<sup>-1</sup>) and T is the temperature as absolute temperature in Kelvin (K). CO<sub>2</sub> efflux and CH<sub>4</sub> oxidation rate are presented as amount of C (ng) per gram soil (dry weight) per hour. Net flux rates were derived by subtraction of the respective values in blanks from the gross values, to correct for potential gas losses through leakage and sampling removal.

The amount of new C derived from labeled <sup>13</sup>CH<sub>4</sub> that was incorporated into incubated soil was calculated using a 2-pool isotope mixing model:

$$\text{NewC} - \text{CH}_4 = \frac{AT\%^{13}C_{AI} - AT\%^{13}C_{BI}}{AT\%^{13}C_{CH4} - AT\%^{13}C_{BI}} \times 100\% \quad (2)$$

where AT%<sup>13</sup>C<sub>AI</sub> is the atom percentage of <sup>13</sup>C isotope in peat soil after incubation, AT%<sup>13</sup>C<sub>BI</sub> is the initial atom percentage of <sup>13</sup>C in peat soil, and AT%<sup>13</sup>C<sub>CH4</sub> is the atom percentage of <sup>13</sup>C of the added CH<sub>4</sub>. All values can be found in the [Supplementary Information \(Table 1S\)](#).

All the measured and calculated parameters were statistically analyzed with R-Studio, a free and open-source integrated development environment for R (a programming language for statistical computing and graphics). All data presented are mean values from three replications (±SE). ANOVA was used for estimating significant differences (p < 0.05) either between different depths within one microform of each treatment or between different microforms of each depth and treatment. All the coefficients of significance (p-values) are presented in the [Supplementary Materials \(Tables S2–S10\)](#). Main and interaction effects between microforms, depths and treatments were tested with the two-way ANOVA ([Tables S11–S13](#)).

## 3. Results

### 3.1. C, N content and C-to-N ratio in peat profile of studied microforms

The total organic carbon content (C) was similar in all microforms at the same depths and the content significantly increased with depth ([Table 1](#)). In contrast, total nitrogen (N) content was significantly lower for hummocks at all depths as compared to both lawns and hollows ([Table 1](#)). The C:N ratio strongly decreased with depth, being 2.5–3 times lower in deeper horizons (30–37) as compared with top 15 cm (85–98). A significantly higher C:N ratio was found in hummocks at 50 and 200 cm depths as compared to lawns and hollows ([Table 1](#)).

### 3.2. CO<sub>2</sub> production from soil with and without nitrate and sulfate amendment

Average CO<sub>2</sub> production rate in soil without NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>

**Table 1**

Carbon (C), nitrogen (N) contents ( $\text{g kg}^{-1}$ ) and the C-to-N ratio (C:N) in peat samples of three microform types (hummocks, lawns, hollows) at three depths (15, 50 and 200 cm). Values are averages of three replicates ( $\pm$ SE). All values were significantly different between the three depths of each microform ( $p < 0.05$ ).

Depth (cm)	Hummocks			Lawns			Hollows		
	C	N	C:N	C	N	C:N	C	N	C:N
15	448.5 (3.1)	4.6 (0.4)*	98.8 (9.9)	451.7 (5.6)	5.3 (0.3)	85.4 (3.9)	448.9 (0.2)	5.3 (0.2)	85.4 (3.7)
50	492.7 (6.2)	13.3 (0.2)*	37.1 (0.3)*	502.7 (3.3)	15.3 (0.1)	32.8 (0.1)	508.3 (0.3)	15.1 (0.3)	33.8 (0.7)
200	562.3 (1.9)	16.9 (0.0)*	33.3 (0.0)*	566.1 (5.8)	19.1 (0.2)	29.6 (0.2)	568.3 (6.2)	19.0 (0.2)	29.9 (0.1)

\*Significantly different values between microforms ( $p < 0.05$ ).

additions revealed contrasting patterns between microforms with depth (Fig. 1, blue color). Thus,  $\text{CO}_2$  efflux for hollows showed the expected pattern: the highest efflux was in the topsoil and significantly decreased with depth (Fig. 1c). In contrast, the

highest  $\text{CO}_2$  efflux (among all microforms) was measured at 50 cm in lawns, whereas the top 15 cm and 200 cm did not differ between each other (Fig. 1b). The average  $\text{CO}_2$  flux in hummocks showed a similar pattern to lawns. However,  $\text{CO}_2$  efflux at depth horizons between hummocks and lawns was not significantly different (Fig. 1a).

For all microforms and at all depths, amendment with  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  decreased  $\text{CO}_2$  efflux by a factor of 1.3–2.9 relative to soil without  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  addition (Fig. 1). The largest suppression was achieved with the addition of  $\text{SO}_4^{2-}$  in hummocks and lawns, whereas  $\text{NO}_3^-$  decreased  $\text{CO}_2$  flux most strongly for hollows. Except for  $\text{NO}_3^-$  addition in hollows, the highest  $\text{CO}_2$  efflux was observed at 50 cm depth in all amended microforms (Fig. 1). In the top and bottom horizons, the difference in  $\text{CO}_2$  effluxes between  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  additions was negligible for all microforms. Main and interaction effects between microforms, depths and treatments as related to  $\text{CO}_2$  production were highly significant, except the “depth-treatment” interaction effect (Table S11).

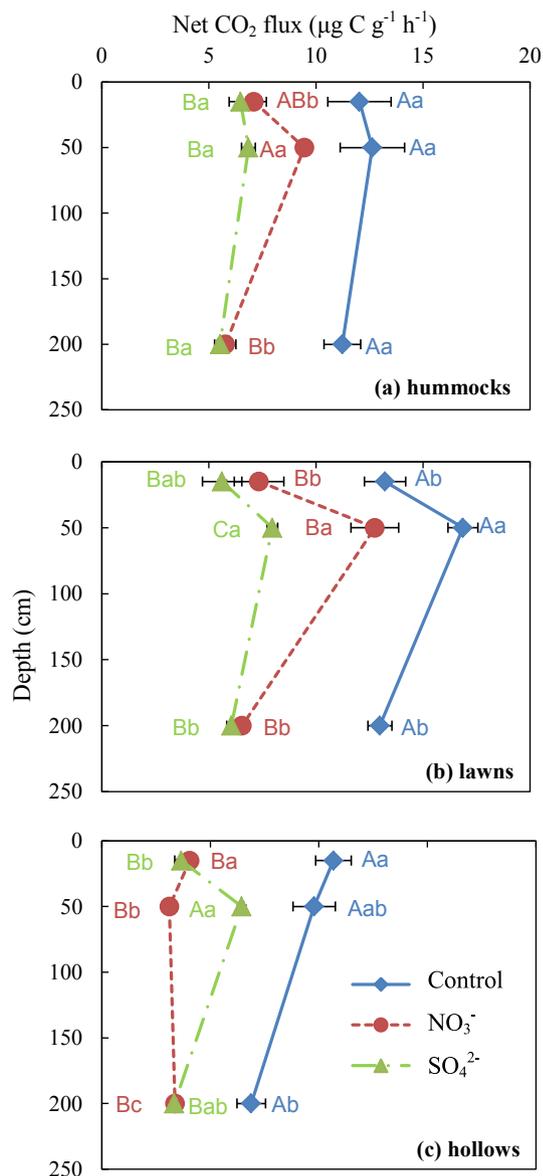
### 3.3. $\text{CH}_4$ oxidation from soil with and without nitrate and sulfate amendment

The most intensive  $\text{CH}_4$  oxidation ( $1.5\text{--}1.7 \mu\text{g C g}^{-1} \text{h}^{-1}$ ) was recorded for the topsoil from lawns and hollows (Fig. 2b, c, blue color, shown as negative values representing the decrease of added  $\text{CH}_4$ ). Remarkably, the  $\text{CH}_4$  oxidation in topsoil from hummocks was ca. 10 times lower, compared with the other two microforms (Fig. 2a). For all microforms, the oxidation dropped significantly with depth, approaching near-zero values at 200 cm. There was no difference in  $\text{CH}_4$  oxidation at each of 50 and 200 cm between the three microforms (Fig. 2).

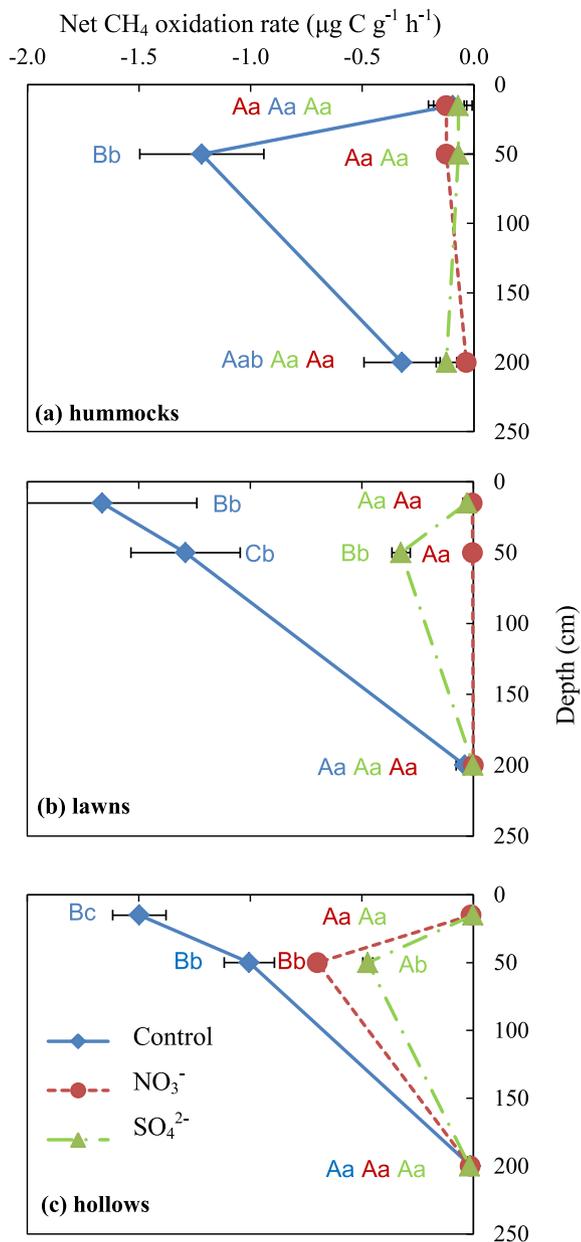
Similar to the  $\text{CO}_2$  response, the addition of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  decreased  $\text{CH}_4$  oxidation in all microforms relative to soil without addition. Though, this effect was only observed at 15 cm (lawns and hollows) and 50 cm depths (all microforms) (Fig. 2). There was no significant effect of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  amendments in the topsoil of hummocks and at 200 cm depth for any microforms as compared to the unamended soil. The suppressing effect of  $\text{NO}_3^-$  was more pronounced in lawns (Fig. 2b), but  $\text{SO}_4^{2-}$  caused higher suppression in hollows (Fig. 2c), whereas  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  both caused a similar decrease in  $\text{CH}_4$  oxidation for hummocks (Fig. 2a). The main effects of microform type, peat depth and treatment with  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  on  $\text{CH}_4$  oxidation, as well as the interaction effects between parameters were significant at  $P < 0.05$  level (Table S12).

### 3.4. $\text{N}_2\text{O}$ production from soil with and without nitrate and sulfate amendment

The  $\text{N}_2\text{O}$  production rate in peat soil without added  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  was 4–5 orders of magnitude lower than the release of C as  $\text{CO}_2$  ( $0.1\text{--}1.2 \text{ ng N g}^{-1} \text{h}^{-1}$ ). In two cases – hummocks at 15 cm and hollows at 50 cm – a negative flux (uptake) was recorded

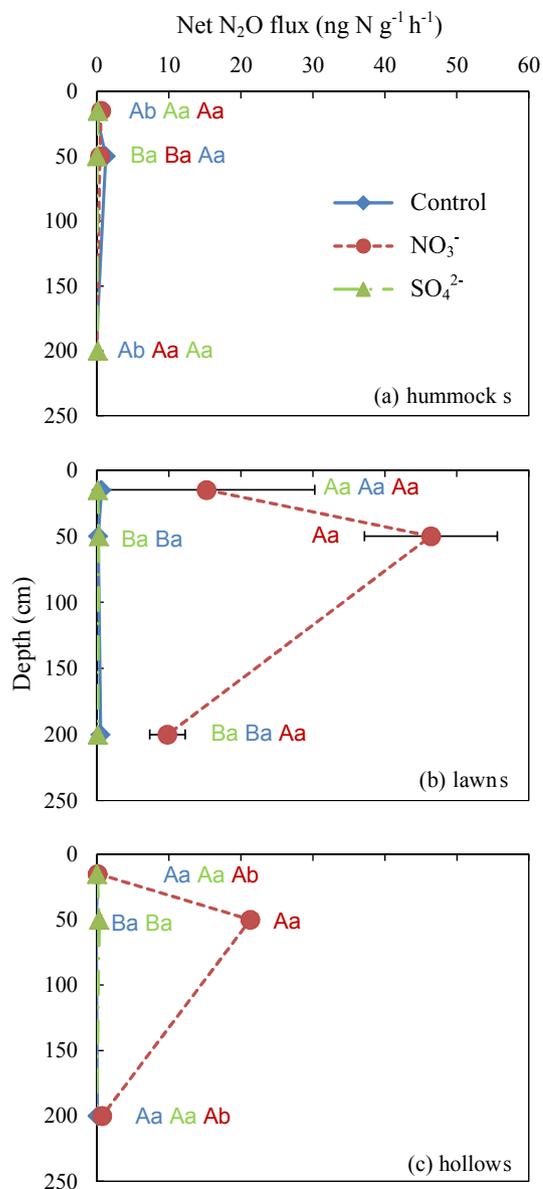


**Fig. 1.** Net  $\text{CO}_2$  flux (per gram (g) of peat dry weight) for: a) hummocks, b) lawns and c) hollows with and without addition of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ . Values are averages of three replicates over the incubation period ( $\pm$ SE). Values followed by different letters are significantly different between the three treatments of the same depth of each microform (uppercase letters) and between the three depths under each treatment within the same microform (lowercase letters) at  $P \leq 0.05$ .



**Fig. 2.** Net CH<sub>4</sub> oxidation (per gram (g) of peat dry weight) for: a) hummocks, b) lawns and c) hollows with and without addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>. Values are averages of three replicates over the incubation period (±SE). Values followed by different letters are significantly different between the three treatments of the same depth of each microform (uppercase letters) and between the three depths under each treatment within the same microform (lowercase letters) at P ≤ 0.05.

(Fig. 3). This indicated an overall low natural N<sub>2</sub>O production potential for the studied boreal peatland. Addition of SO<sub>4</sub><sup>2-</sup> did not change the pattern observed in the unamended soil, whereas NO<sub>3</sub><sup>-</sup> amendment resulted in the N<sub>2</sub>O release. Thus, the highest rate of 46.4 ± 9.1 ng N g<sup>-1</sup> h<sup>-1</sup> was measured for lawns at 50 cm depth (Fig. 3b), followed by a 2-fold lower rate (21.3 ± 0.7 ng N g<sup>-1</sup> h<sup>-1</sup>) for hollows at the same depth (Fig. 3c). The rate of N<sub>2</sub>O production was significantly increased after NO<sub>3</sub><sup>-</sup> amendment for lawns at 200 cm, whereas in the topsoil of lawns, as well as in the top and bottom horizons of hollows, the N<sub>2</sub>O production did not significantly differ from that of the unamended soil. The addition of NO<sub>3</sub><sup>-</sup> had no effect in soil of hummocks from any depth (Fig. 3a). This also reflected the insignificant interaction effect between

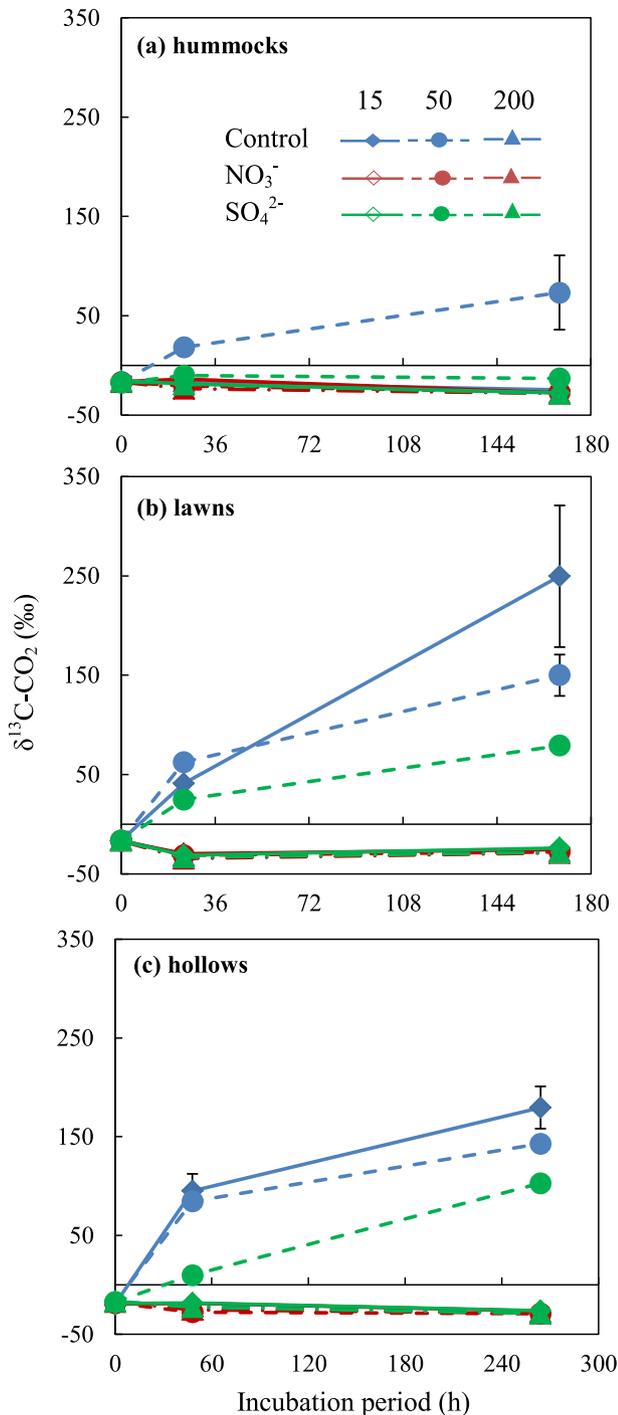


**Fig. 3.** Net N<sub>2</sub>O flux (per gram (g) of peat dry weight) for: a) hummocks, b) lawns and c) hollows with and without addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>. Values are averages of three replicates over the incubation period (±SE). Values followed by different letters are significantly different between the three treatments of the same depth of each microform (uppercase letters) and between the three depths under each treatment within the same microform (lowercase letters) at P ≤ 0.05.

microform type and depth (Table S13), whereas other main effects of microform type, depth and treatment with other respective interaction effects were highly significant (p < 0.01).

### 3.5. CO<sub>2</sub> produced from oxidation of labeled CH<sub>4</sub>

Application of <sup>13</sup>C-labeled CH<sub>4</sub> allowed the tracing of released <sup>13</sup>C-CO<sub>2</sub> (Fig. 4), and a comparison with the mass-based CH<sub>4</sub> oxidation (Fig. 2). In general, the intensity of CH<sub>4</sub> oxidation corresponded to the <sup>13</sup>C enrichment of CO<sub>2</sub>. Thus, the highest CH<sub>4</sub> oxidation rate measured in topsoil of lawns under the soil without NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> addition (Fig. 2b) corresponded to the most enriched δ<sup>13</sup>C-CO<sub>2</sub> (up to 170‰) detected in the same treatment, microform and depth (Fig. 4b). CH<sub>4</sub> oxidation in hummocks (50 cm, control),



**Fig. 4.** Delta $^{13}\text{C}$  of  $\text{CO}_2$  values for: a) hummocks, b) lawns and c) hollows with and without addition of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ . Diamonds, circles and triangles represent three depth horizons (15, 50 and 200 cm), respectively. Values are averages of three replicates ( $\pm$ SE).

lawns (50 cm, control and  $\text{SO}_4^{2-}$ ) and hollows (15 cm, control; 50 cm, control and  $\text{SO}_4^{2-}$ ) corresponded well with the respective dynamics of  $\delta^{13}\text{C}-\text{CO}_2$  values (Fig. 4). However, for soil from 50 cm depth of hollows with  $\text{NO}_3^-$  addition (Fig. 2c) the substantial rate of  $\text{CH}_4$  oxidation did not result in production of labeled  $\text{CO}_2$  (i.e.  $\text{CO}_2$  originating from  $\text{CH}_4$  oxidation),  $\delta^{13}\text{C}-\text{CO}_2$  values were close to the natural abundance of  $\text{CO}_2$  efflux (Fig. 4c). For other samples, their much less intensive or near-zero  $\text{CH}_4$  oxidation rates revealed no

clear  $^{13}\text{C}-\text{CO}_2$  enrichment, which was most probably diluted and masked by the background  $\text{CO}_2$  derived from SOM decomposition. Moreover,  $\delta^{13}\text{C}$  values of  $\text{CO}_2$  decreased (became more negative) during incubation and this decrease was more pronounced with  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  addition, indicating an increasing contribution of native SOM decomposition with time.

### 3.6. Relationship between microbial biomass carbon and new $^{13}\text{C}-\text{CH}_4$ -derived carbon in soil

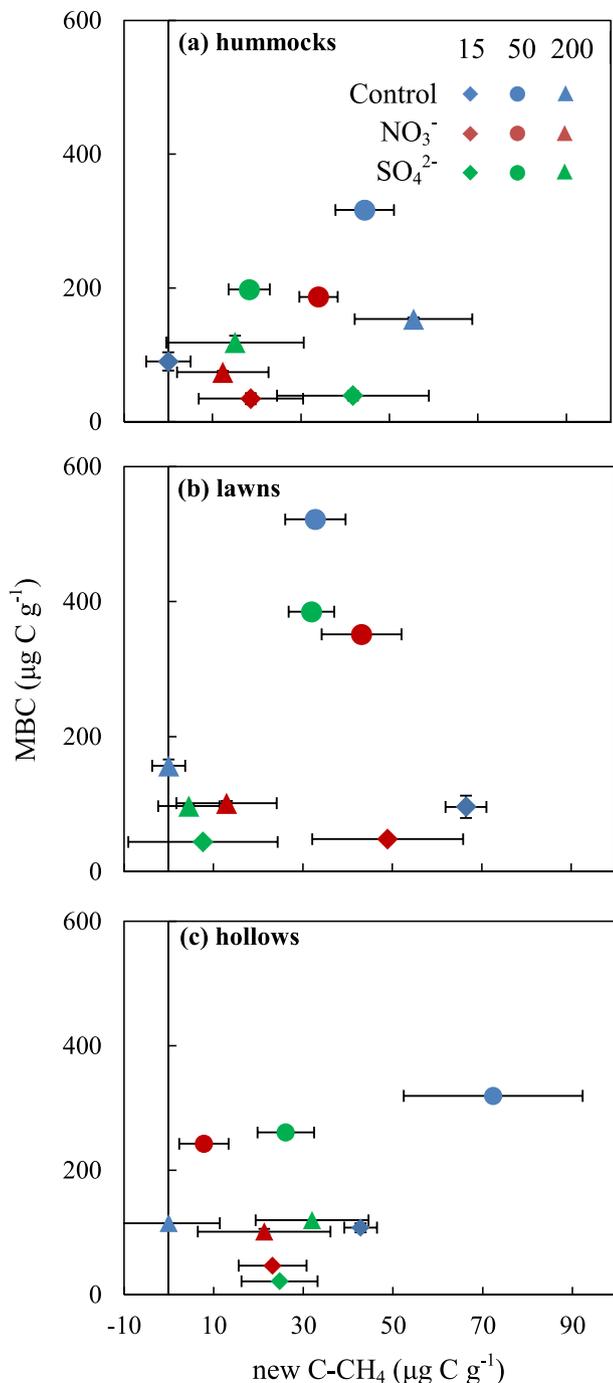
Microbial biomass C (MBC) content was highest at 50 cm depth for all microforms (Fig. 5). Between microforms, lawns contained the highest MBC (Fig. 5b). Amendment with  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  resulted in an overall decrease of the DNA-extractable C relative to the unamended soil (Fig. 5). Thus, for hummocks, MBC was 41–62% lower with  $\text{NO}_3^-$  and 23–57% lower with  $\text{SO}_4^{2-}$  as compared with the unamended soil, depending on depth. For lawns, the decrease was 33–50% and 26–54% with  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  amendment, respectively. Addition of  $\text{NO}_3^-$  in hollows decreased the MBC content by 12–57% and the decrease due to  $\text{SO}_4^{2-}$  amendment was 0–81% between all depths (Fig. 5).

The  $^{13}\text{C}$  enrichment of total OM before and after the incubation experiment (Table 1S) and the portion of new  $^{13}\text{C}-\text{CH}_4$  incorporation into OM showed new C share from the MBC (Fig. 5, X-axis). As the MBC constitutes a part of the total OM in soil, the new C was derived from microorganisms consuming  $\text{CH}_4$  (methanotrophs). The relationship between total MBC and the amount of new C roughly demonstrates the relative contribution of methanotrophs to the total microbial biomass in various microforms (Fig. 6).

The amount of new  $\text{CH}_4$ -derived C for hummocks without addition of anions increased with depth (15 < 50 < 200 cm). The amendment with  $\text{NO}_3^-$  did not reveal particular trend (50 > 15 > 200 cm), whereas a decreasing trend was detected with depth with  $\text{SO}_4^{2-}$  addition (15 > 50 > 200 cm) (Fig. 5a). In soil after the addition of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ , the ratio of new  $^{13}\text{C}-\text{CH}_4$ -derived C to total MBC was highest in the top horizon, indicating a greater presence of methanotrophs or their higher activity as compared with deeper soil horizons (Fig. 6, hummocks). However, this was not true in soil without addition, for which no measurable differences in new C incorporation and microbial biomass (negative values) were detected between different depths (Fig. 6, hummocks).

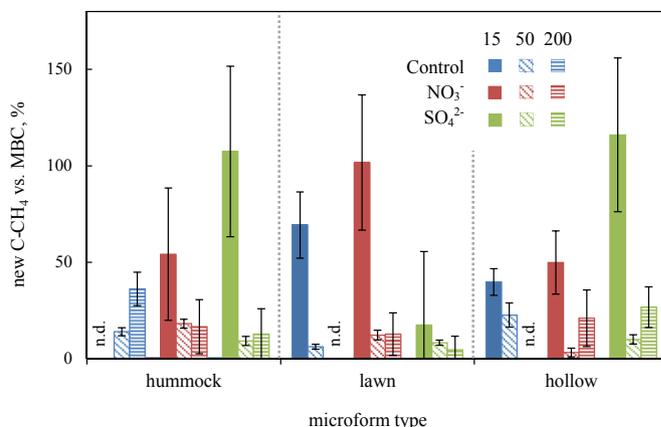
In contrast to hummocks, the observed incorporation of new  $\text{CH}_4$ -derived C in soil from lawns without  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  amendment decreased with depth: 15 > 50 > 200 cm. The same pattern was detected after  $\text{NO}_3^-$  addition. No clear depth effect was recorded after  $\text{SO}_4^{2-}$  addition: 50 > 15 > 200 cm (Fig. 5b). The highest ratio of new  $^{13}\text{C}-\text{CH}_4$ -derived C to total MBC was detected in the topsoil (Fig. 6, lawns).  $\text{NO}_3^-$  amendment promoted incorporation of new C which comprised up to 100% of MBC, whereas  $\text{SO}_4^{2-}$  addition did not affect the ratio of new C to MBC as compared to other treatments at 15 cm depth. The relatively low ratio of new C to MBC in deeper horizons in comparison to the topsoil may result from several factors: (i) “dilution” by abundant unlabelled MBC e.g. at 50 cm depth, Fig. 5b, (ii) much lower activity of methanotrophs at the very depth (200 cm) and/or (iii) low level of electron acceptors in soil microzones of the bottom horizon, where  $\text{O}_2$  diffusion is restricted due to the overall high moisture content.

For hollows, the highest amount of new C (more than  $70 \mu\text{g g}^{-1}$ ) was observed for unamended soil from 50 cm depth, followed by 15 cm depth, whereas no incorporation of new C was detected for 200 cm depth under the same treatment (Fig. 5c).  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  amendments decreased the amount of new C at 15 and 50 cm as compared to the control. In contrast, at 200 cm,



**Fig. 5.** Amount of microbial biomass carbon (MBC) and amount of new <sup>13</sup>CH<sub>4</sub>-derived carbon (per gram (g) of peat dry weight) for: a) hummocks, b) lawns and c) hollows with and without addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>. Diamonds, circles and triangles represent three depth horizons (15, 50 and 200 cm), respectively. Values are averages of three replicates ( $\pm$ SE).

the new CH<sub>4</sub>-derived C increased after the addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> (Fig. 5c). Similar to hummocks and lawns, the ratio of new C to MBC in hollows was higher in the topsoil of the control treatment as compared to 50 and 200 cm and further increased after the addition of NO<sub>3</sub><sup>-</sup> and especially of SO<sub>4</sub><sup>2-</sup> (Fig. 6, hollow). There was no difference in the ratio of new C to MBC between soils with and without NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> addition for 50 and 200 cm-deep horizons.



**Fig. 6.** Proportion between new <sup>13</sup>CH<sub>4</sub>-derived carbon and total MBC content for hummocks, lawns and hollows in three depth horizons. The source of the values is data from Fig. 5 (new <sup>13</sup>CH<sub>4</sub>-C/MBC\*100%). Blue color corresponds to the control treatment (without NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>), red to NO<sub>3</sub><sup>-</sup> and green to SO<sub>4</sub><sup>2-</sup> amendment. Uniformly filled bars and bars with downward diagonal and horizontal patterns represent the three depth horizons (15, 50 and 200 cm), respectively. Values are averages of three replicates ( $\pm$ SE). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

#### 4. Discussion

##### 4.1. Effect of microform types and soil depth on CO<sub>2</sub> efflux, CH<sub>4</sub> oxidation and N<sub>2</sub>O production (hypothesis 1 and 2)

The lowest CO<sub>2</sub> efflux was attributed to the hollows (Fig. 1, blue color), which naturally experience stronger anaerobic conditions due to the higher water table (WT) level as compared to lawns and hummocks (Becker et al., 2008; Dorodnikov et al., 2013). However, the pattern was not clearly linked to O<sub>2</sub> availability: the overall CO<sub>2</sub> efflux of the driest microform – hummocks – did not significantly differ from the wettest hollows (data not shown). *In vitro* conditions strongly decreased limitations of fluctuating natural environmental factors, thereby revealing the differences in the constituent soil properties. Therefore, under controlled conditions, i) the type and abundance of decomposers (Basiliko et al., 2007; Strakova et al., 2012), and consequently ii) the quality and quantity of the substrate they are decomposing (Moore and Dalva, 1997; Yavitt et al., 2000; Blagodatskaya et al., 2010; Strakova et al., 2012) become the main determinants of CO<sub>2</sub> fluxes among the microforms. Thus, the MBC explained ca. 21% of the variation in all measured CO<sub>2</sub> fluxes (Fig. S1a). The highest correlation ( $R^2 = 0.75$ ) was observed in soil of the top 15 cm horizon, with and without NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> (Fig. S1d). Within the total microbial population, part of the CO<sub>2</sub> efflux was also connected with the activity of methanotrophs. However, we observed a negative relationship (Fig. S2a): samples showing higher CH<sub>4</sub> oxidation did not express increased CO<sub>2</sub> efflux. Between microforms, such a pattern was the most pronounced in hummocks (explaining 49% of the measured variation, Fig. S2b), and between depths – in the topsoil of microforms (explaining 68% of the variation, Fig. S2d). Therefore, measured soil CO<sub>2</sub> efflux conditionally confirmed the hypothesis of more intensive OM decomposition in soils better adapted to aerobic conditions (Hypothesis 1) and was related to MBC. However, seemingly other environmental factors as oxygen availability controlled MBC distribution and related CO<sub>2</sub> fluxes among the microforms (see below). Decrease in CO<sub>2</sub> production with the microforms' depth was hypothesized to be linked with SOM turnover through the depletion of fresh plant-derived C inputs available for decomposition and strict anaerobic conditions. Aerobic decomposition in the topsoil

typically reduces the quality of litter entering the deeper horizons (Strakova et al., 2012), thereby affecting CO<sub>2</sub> production (Saarnio et al., 1998). Since the organic C content increased with depth (Table 1), the occurrence of the lowest CO<sub>2</sub> production at the bottom of the profile for all microforms (Fig. 1) can be attributed to (i) the properties of the substrate for decomposition and (ii) properties of the soil microbial biomass (e.g. strict anaerobes at the bottom soil horizons could not tolerate the increased O<sub>2</sub> availability in the incubation experiment). Remarkably, decomposability of the organic substrate in the studied soil could not be described by the commonly used C:N parameter, because the significant decrease of C:N with depth (Table 1) was uncoupled from the rate of CO<sub>2</sub> efflux (Fig. 1). In contrast to the hypothesized patterns (Hypothesis 2), in hummocks and especially in lawns, the highest CO<sub>2</sub> efflux corresponded to 50 cm depth and not to the topsoil. This highest CO<sub>2</sub> efflux was related to the largest microbial biomass content at this depth (Fig. 5, Y-axis, blue color) and the positive correlation between both parameters explained 34% of their variation (Fig. S1d).

While the natural aeration gradient is partly responsible for differences in CO<sub>2</sub> production from the microforms, this gradient was not associated with the expected CH<sub>4</sub> oxidation patterns. Namely, there was higher CH<sub>4</sub> oxidation where natural aeration is lower, e.g. in lawns and hollows rather than hummocks (Fig. 2, blue color). Several authors have reported positive correlation between CH<sub>4</sub> fluxes and CH<sub>4</sub> oxidation rates (Basiliko et al., 2007; Hornibrook et al., 2009). This may suggest that CH<sub>4</sub> oxidation is substrate (CH<sub>4</sub>)-dependent rather than limited by the availability of O<sub>2</sub> (Sundh et al., 1994; Saarnio et al., 1997). Current data demonstrate that in the topsoil of hummocks, neither mass-based oxidation (Fig. 2a) nor the incorporation of new <sup>13</sup>C-CH<sub>4</sub> to the total OM (Fig. 5a), nor its ratio to MBC (Fig. 6), suggest high oxidation potential of the aerobic zone of hummocks. Instead, low *in situ* CH<sub>4</sub> fluxes from these microforms are presumably related to their low methanogenic potential (Saarnio et al., 1997).

Interestingly, indirect evidence of methanotrophic activity, shown as a relationship between the amount of new C derived from CH<sub>4</sub> and the CO<sub>2</sub> flux (Fig. S3), demonstrated the highest correlation ( $R^2 = 0.64$ ) at 50 cm depth for all microforms (Fig. S3d). The correlation between new CH<sub>4</sub>-derived C incorporated into OM during incubation and CH<sub>4</sub> oxidation was surprisingly weak, explaining only 9% of the overall variation between the two variables (Fig. S4a). However, the estimated negative relationship between new C and CH<sub>4</sub> oxidation generally counteracted the observed production of labeled CO<sub>2</sub> (Fig. 4) and a positive correlation between MBC content and new C (Fig. S5a). Relatively weak correlation between new C and CH<sub>4</sub> oxidation, as well as between new C and MBC, was observed because the oxidized <sup>13</sup>CH<sub>4</sub> was either 1) strongly diluted in the bulk of organic matter or 2) was rapidly turned over and released as <sup>13</sup>CO<sub>2</sub> without substantial incorporation into SOM.

The production of N<sub>2</sub>O in the microforms and with the depth of the unamended soil (reference, natural conditions) appeared to be not testable with the proposed hypotheses as the majority of fluxes showed close to “zero” rates (Fig. 3, blue color). Markedly low N<sub>2</sub>O flux was most probably related to the analytical approach as the substantial dilution of the headspace gas samples with N<sub>2</sub> was required for the equipment. Still, the occurrence of microorganisms responsible for nitrification/denitrification processes was indirectly revealed in two of the three microforms (lawns and hollows) after the addition of NO<sub>3</sub><sup>-</sup> (Fig. 3b, c).

Summarizing, the expected increase in the GHG production with the naturally established aeration zone of microforms (Hypothesis 1) was conditionally approved for CO<sub>2</sub> fluxes (hollows < hummocks ≤ lawns), whereas CH<sub>4</sub> oxidation potential was the lowest in the most aerated hummocks followed by hollows and lawns. With depth, neither CO<sub>2</sub> fluxes, nor CH<sub>4</sub> oxidation and

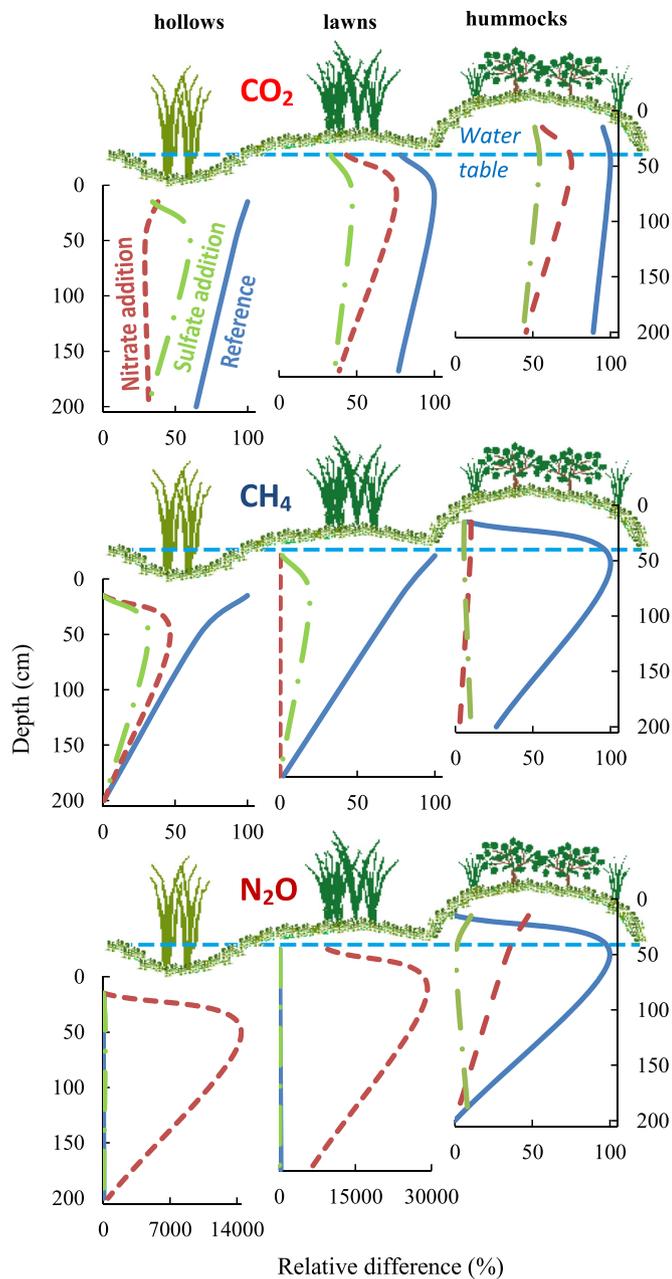
N<sub>2</sub>O production revealed the expected gradual decrease (Hypothesis 2 rejected). Instead, the highest values corresponded to the 50 cm peat horizon. Therefore, the *in situ* differences in oxygen availability among the studied microforms appeared to be less significant factor for GHG dynamics, whereas other constituent soil properties, such as the microbial biomass content, were responsible for GHG dynamics.

#### 4.2. Effects of nitrate and sulfate addition on CO<sub>2</sub> efflux, CH<sub>4</sub> oxidation and N<sub>2</sub>O production (hypothesis 3)

In *in vitro* incubation, there is no competition with plants and the soil microorganisms have access to the whole amount of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> added. The properties of the microbial community inherited from *in situ* conditions revealed differences between microforms and depths. In general, addition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> suppressed CO<sub>2</sub> production, as compared to soil without addition, for all microforms (Fig. 1; significant effect of the treatment, Table S11). These results do not support the hypothesis of increased CO<sub>2</sub> fluxes due to the nutritional effect of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> (Hypothesis 3). Suppression due to NO<sub>3</sub><sup>-</sup> addition was less pronounced as compared to SO<sub>4</sub><sup>2-</sup> amendment (change in MBC under NO<sub>3</sub><sup>-</sup> treatment could explain just 4% of the variation in CO<sub>2</sub> flux vs. 29% under SO<sub>4</sub><sup>2-</sup> addition, Fig. S1c). This confirms that NO<sub>3</sub><sup>-</sup> participated in processes related to the broader functionality of the microbial community, not just the decomposition of SOM.

CH<sub>4</sub> oxidation was suppressed under NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> addition (Fig. 7; treatment main effects, Table 12S). On average 40% lower CH<sub>4</sub> flux was measured in soil from several northern peatlands after the addition of SO<sub>4</sub><sup>2-</sup>, whereas around 90% reduction was measured with added NO<sub>3</sub><sup>-</sup> in comparison to reference soil (Dettling et al., 2006). However, it remained unclear whether the decrease resulted from the CH<sub>4</sub> oxidation or suppression of methanogenesis. The latter phenomenon was explained, in the case of NO<sub>3</sub><sup>-</sup> amendment, by the occurrence of denitrification intermediates such as nitrite (NO<sub>2</sub><sup>-</sup>), nitrogen dioxide (NO<sub>2</sub>), nitric oxide (NO), which are known to have suppressing effects on CH<sub>4</sub> production (Chen and Lin, 1993; Clarens et al., 1998; Eriksson et al., 2010). Despite decreased CH<sub>4</sub> oxidation in amended soil, the proportion between new C incorporation to SOM and MBC content showed the highest values (up to 100%) under both NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, especially in the topsoil (Fig. 6, green and red color). This finding may indicate the following processes: (i) increased substrate use efficiency by methanotrophs in the topsoil as compared with deeper soil horizons, when <sup>13</sup>C-CH<sub>4</sub> retains in microbial cells instead of being quickly metabolized and respired (thus, no pronounced <sup>13</sup>C-CO<sub>2</sub> enrichment was detected (Fig. 4)); (ii) relative suppression of methanotrophs due to NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> amendment was not as intensive as of other microbial groups because total MBC decreased under the respective treatments (Fig. 5); (iii) predation of methanotrophs by other microorganisms or animals distributed the <sup>13</sup>C label within the soil but diluted <sup>13</sup>CO<sub>2</sub> with <sup>12</sup>CO<sub>2</sub> from other metabolic processes. These mechanisms should be tested in separate experiments with the determination of microbial community structure. Although the decreased CH<sub>4</sub> oxidation was assumed to be due to the more energetically favorable processes, e.g. nitrification/denitrification (Hypothesis 3 was conditionally supported), a positive correlation between CH<sub>4</sub> oxidation and new C in SOM was detected under NO<sub>3</sub><sup>-</sup> amendment (Fig. S4c, relationship explained 57% of the observed variation). This may suggest that CH<sub>4</sub> oxidation was not fully outcompeted by denitrification or other processes, e.g. anaerobic oxidation of methane (AOM) may occur (see below).

Similar to soil without additions, SO<sub>4</sub><sup>2-</sup> had no significant effect on N<sub>2</sub>O production and the fluxes were close to zero for all microforms and depths (Fig. 3, green color) suggesting that S was not a



**Fig. 7.** Effects of sulfate (green dashed-dotted lines) and nitrate (red dashed lines) additions on fluxes of CO<sub>2</sub> (top), CH<sub>4</sub> (middle) and N<sub>2</sub>O (bottom) for hummocks, lawns and hollows at increasing peat depth as compared to a control treatment without addition (reference, blue line). Effects are shown as relative difference between control treatment (100%) and respective sulfate or nitrate treatments. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

limiting nutrient for nitrifying/denitrifying microorganisms. As expected, NO<sub>3</sub><sup>-</sup> amendment increased N<sub>2</sub>O production in comparison to unamended soil (by 15,000–30,000%), but the effect was observed only for lawns at 50 and 200 cm and for hollows at 50 cm (Fig. 7). Surprisingly, no effect of NO<sub>3</sub><sup>-</sup> addition on N<sub>2</sub>O was measured for hummocks at any depth (Fig. 3a). Such a contrasting pattern between microforms and depths could be related to strong variations in microbial community structure (Kotiahho et al., 2013; Deng et al., 2014) and multiple factors may affect the occurrence and/or activity of nitrifying/denitrifying microorganisms.

It is important to note, that in hollows at 50 cm both N<sub>2</sub>O

production and CH<sub>4</sub> oxidation were observed under NO<sub>3</sub><sup>-</sup> addition (Figs. 2c and 3c). Despite the aerobic incubation, the water content of the peat soil reached 95% by weight, therefore microzones with anaerobic conditions may have persisted in the samples during the experiment. Hence, the CH<sub>4</sub> oxidation could also happen via AOM processes (Smemo and Yavitt, 2011). AOM based on NO<sub>3</sub><sup>-</sup> as an alternative electron acceptor to oxygen was predicted to occur in peatlands, because reduction of N oxides provides sufficient free energy to fuel CH<sub>4</sub> oxidation (Smemo and Yavitt, 2011). The potential of added NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> as electron acceptors for AOM in the studied soil would have to be tested under anaerobic conditions in a separate experiment.

## 5. Conclusions

The undertaken measurements of CO<sub>2</sub> and N<sub>2</sub>O production, CH<sub>4</sub> oxidation, microbial biomass content and incorporation of <sup>13</sup>CH<sub>4</sub>-derived C into peat soil samples with and without amendment of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>, lead to the following conclusions:

- **Effects of microforms:** CO<sub>2</sub> efflux decreased in the order lawns ≥ hummocks > hollows (Hypothesis 1 conditionally accepted), however CH<sub>4</sub> oxidation did not follow the naturally established aerobic zone of a microform type and increase from hollows to lawns and to hummocks. In contrast to oxygen availability, MBC content was the key factor controlling the processes in the microforms. Patterns of N<sub>2</sub>O production were not testable with the Hypothesis 1 due to low fluxes in peat under natural conditions.
- **Effects of depth:** CO<sub>2</sub> efflux, CH<sub>4</sub> oxidation and N<sub>2</sub>O fluxes did not confirm the hypothesized descend with depth due to *in situ* decreasing availability of oxygen and fresh plant-derived deposits (Hypothesis 2). Remarkably, the highest GHG fluxes as well as MBC content were observed at 50 cm depth below all microforms (Fig. 7).
- **Effects of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> amendments:** CO<sub>2</sub> efflux decreased under both NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> amendments as compared to soil without addition, for all microforms and depths, following the decrease in the microbial DNA-extractable C. This rejected the nutritional aspect in the Hypothesis 3. Contrastingly, the CH<sub>4</sub> oxidation was retarded by 20–94% after the amendment and did not generally coexist with the N<sub>2</sub>O production, hereby supporting the preferential process aspect in the Hypothesis 3.

## 6. Outlook

In a broader ecological view, nitrate and sulfate deposition may suppress CO<sub>2</sub> efflux, which is positive for GHG mitigation and climate change. On the other hand, CH<sub>4</sub> oxidative potential could be suppressed either. This would lead to more intensive CH<sub>4</sub> release to the atmosphere, presumably due to the CH<sub>4</sub> produced from older C stored in the system, thereby compensating positive effect of reduced CO<sub>2</sub> production. Moreover, increased NO<sub>3</sub><sup>-</sup> deposition would stimulate N<sub>2</sub>O formation and promote contribution of this very potent GHG to the atmosphere. Taken together, human-induced deposition of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> may suppress CO<sub>2</sub> emissions from and CH<sub>4</sub> oxidation by boreal oligotrophic mires especially under the conditions of deposition increase. Therefore, the deposition of inorganic compounds is strongly important to be considered in the estimation of ecosystem C and N balances.

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## Appendix A. Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.soilbio.2016.06.018>.

## References

- Alm, J., Saarnio, S., Nykanen, H., Silvola, J., Martikainen, J.P., 1999. Winter  $\text{CO}_2$ ,  $\text{CH}_4$  and  $\text{N}_2\text{O}$  fluxes on some natural and drained boreal peatlands. *Biogeochemistry* 44, 163–186.
- Basiliko, N., Blodau, C., Roehm, C., Bentson, P., Moore, T.R., 2007. Regulation of decomposition and methane dynamics across natural, commercially mined, and restored Northern Peatlands. *Ecosystems* 10, 1148–1165.
- Becker, T., Kutzbach, L., Forbrich, I., Schneider, J., Jager, D., Thees, B., Wilmking, M., 2008. Do we miss the hot spots? the use of very high-resolution aerial photographs to quantify carbon fluxes in peatlands. *Biogeosciences* 5, 1387–1393.
- Blagodatskaya, E., Blagodatsky, S., Anderson, T.-H., Kuzyakov, Y., 2014. Microbial growth and carbon use efficiency in the rhizosphere and root-free soil. *PLoS One* 9, e93282. <http://dx.doi.org/10.1371/journal.pone.0093282>.
- Blagodatskaya, E., Blagodatsky, S., Dorodnikov, M., Kuzyakov, Y., 2010. Elevated atmospheric  $\text{CO}_2$  increases microbial growth rates in soil: results of three  $\text{CO}_2$  enrichment experiments. *Glob. Change Biol.* 16, 836–848.
- Bodegom, P.M., Stams, A.J.M., 1999. Effect of alternative electron acceptors and temperature on methanogenesis in rice paddy soils. *Chemosphere* 39, 167–182.
- Chen, K.C., Lin, Y.F., 1993. The relationship between denitrifying bacteria and methanogenic bacteria in mixed culture system of acclimated sludges. *Water Resour.* 27, 1749–1759.
- Clarens, M., Bernet, N., Delgenès, J.F., Moletta, R., 1998. Effects of nitrogen oxides and denitrification by *Pseudomonas stutzeri* on acetotrophic methanogenesis by *Methanosarcina mazei*. *Microbiol. Ecol.* 25, 271–276.
- Deng, Y.C., Cui, X.Y., Hernandez, M., Dumont, M.G., 2014. Microbial diversity in hummock and hollow soils of three wetlands on the Qinghai-Tibetan Plateau revealed by 16S rRNA Pyrosequencing. *PLoS One* 9, e103115. <http://dx.doi.org/10.1371/journal.pone.0103115>.
- Dettling, M.D., Yavitt, J.B., Zinder, H.S., 2006. Control of organic carbon mineralization by alternative electron acceptors in four peatlands, Central New York State, USA. *Wetlands* 26, 917–927.
- Dorodnikov, M., Knorr, K.H., Kuzyakov, Y., Wilmking, M., 2011. Plant-mediated  $\text{CH}_4$  transport and contribution of photosynthates to methanogenesis at a boreal mire: a  $^{14}\text{C}$  pulse-labeling study. *Biogeosciences* 8, 2365–2375.
- Dorodnikov, M., Marushchak, M., Biasi, C., Wilmking, M., 2013. Effect of microtopography on isotopic composition of methane in porewater and efflux at a boreal peatland. *Boreal Environ. Res.* 18, 269–279.
- Eriksson, T., Oquist, M.G., Nilsson, M.B., 2010. Production and oxidation of methane in a boreal mire after a decade of increased temperature and nitrogen and sulfur deposition. *Glob. Change Biol.* 16, 2130–2144.
- Gauci, V., Matthews, E., Dise, N., Walter, B., Koch, D., Granberg, G., Vile, M., 2004. Sulfur pollution suppression of the wetland methane source in the 20th and 21st centuries. *Proc. Natl. Acad. Sci. U. S. A.* 101, 12583–12587.
- Goldberg, D.S., Knorr, K.H., Blodau, C., Lischeid, G., Gebauer, G., 2010. Impact of altering the water table height of an acidic fen on  $\text{N}_2\text{O}$  and  $\text{NO}$  fluxes and soil concentrations. *Glob. Change Biol.* 16, 220–233.
- Gorham, E., 1991. Northern peatlands – role in the carbon-cycle and probable responses to climatic warming. *Ecol. Appl.* 1, 182–195.
- Hornibrook, E.R.C., Bowes, H.L., Culbert, A., Gallego-Sala, A.V., 2009. Methanotrophy potential versus methane supply by pore water diffusion in peatlands. *Biogeosciences* 6, 1491–1504.
- IPCC, 2014. In: Core Writing Team, Pachauri, R.K., Meyer, L.A. (Eds.), *Climate Change 2014: Synthesis Report. Contribution of Working Groups I, II and III to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*. IPCC, Geneva, Switzerland, 151pp.
- Kotiaho, M., Fritze, H., Merila, P., Tuomivirta, T., Valiranta, M., Korhola, A., Karofeld, E., Tuittila, E.S., 2013. Actinobacteria community structure in the peat profile of boreal bogs follows a variation in the microtopographical gradient similar to vegetation. *Plant Soil* 369, 103–114.
- Kuzyakov, Y., Domanski, G., 2000. Carbon inputs by plants in soil. *Review. J. Plant Nutr. Soil Sci.* 163, 421–431.
- Lafleur, P.M., McCaughey, H.J., Joiner, D.W., Bartlett, P.A., Jelinski, D.E., 1997. Seasonal trends in energy, water, and carbon dioxide fluxes at a northern boreal wetland. *J. Geophys. Res.* 102, 9–20.
- Lai, D.Y.F., 2009. Methane dynamics in northern peatlands: a review. *Pedosphere* 19, 409–421.
- Marushchak, M.E., Pitkamaki, A., Koponen, H., Biasi, C., Seppala, M., Martikainen, P.J., 2011. Hot spots for nitrous oxide emissions found in different types of permafrost peatlands. *Glob. Change Biol.* 17, 2601–2614.
- Moore, T.R., Dalva, M., 1997. Methane and carbon dioxide exchange potential of peat soils in aerobic and anaerobic laboratory incubations. *Soil Biol. Biochem.* 29, 1157–1164.
- Myhre, G., Shindell, D., Bréon, F.-M., Collins, W., Fuglestedt, J., Huang, J., Koch, D., Lamarque, J.-F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G., Takemura, T., Zhang, H., 2013. Anthropogenic and natural radiative forcing. In: Stocker, T.F., Qin, D., Plattner, G.-K., Tignor, M., Allen, S.K., Boschung, J., Nauels, A., Xia, Y., Bex, V., Midgley, P.M. (Eds.), *Climate Change 2013: the Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*. Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA.
- NOAA (National Oceanic and Atmospheric Administration), 2015. National Centers for Environmental Information. Accessed April 2015. [www.ncei.noaa.gov](http://www.ncei.noaa.gov).
- Regina, K., Nykanen, H., Silvola, J., Martikainen, P.J., 1996. Fluxes of nitrous oxide from boreal peatlands as affected by peatland type, water table level and nitrification capacity. *Biogeochemistry* 35, 401–418.
- Roulet, N.T., 2000. Peatlands, carbon storage, greenhouse gases and the Kyoto protocol: prospects and significance for Canada. *Wetlands* 20, 605–615.
- Saarnio, S., Alm, J., Silvola, J., Lohila, A., Nykänen, H., Martikainen, P.J., 1997. Seasonal variation in  $\text{CH}_4$  emissions and production and oxidation potentials at microsites on an oligotrophic pine fen. *Oecologia* 110, 414–422.
- Saarnio, S., Alm, J., Martikainen, P.J., Silvola, J., 1998. Effects of raised  $\text{CO}_2$  on potential  $\text{CH}_3$  production and oxidation in and  $\text{CH}_3$  emission from a boreal mire. *J. Ecol.* 86, 261–268.
- Segarra, K.E.A., Comerford, C., Slaughter, J., Joye, S., 2013. Impact of electron acceptor availability on the anaerobic oxidation of methane in coastal freshwater and brackish wetland sediments. *Geochimica Cosmochimica Acta* 115, 15–30.
- Smemo, K.A., Yavitt, J.B., 2007. Evidence for anaerobic  $\text{CH}_4$  oxidation in freshwater peatlands. *Geomicrobiol. J.* 24, 583–597.
- Smemo, K.A., Yavitt, J.B., 2011. Anaerobic oxidation of methane: an underappreciated aspect of methane cycling in peatland ecosystems? *Biogeosciences* 8, 779–793.
- Strakova, P., Penttila, T., Laine, J., Laiho, R., 2012. Disentangling direct and indirect effects of water table drawdown on above- and belowground plant litter decomposition: consequences for accumulation of organic matter in boreal peatlands. *Glob. Change Biol.* 18, 322–335.
- Sundh, I., Nilsson, M., Granberg, G., Svensson, B.H., 1994. Depth distribution of microbial production and oxidation of methane in northern boreal peatlands. *Microb. Ecol.* 27, 253–265.
- Sutton-Grier, A.E., Keller, J.K., Koch, R., Gilmour, C., Megonigal, J.P., 2011. Electron donors and acceptors influence anaerobic soil organic matter mineralization in tidal marshes. *Soil Biol. Biochem.* 43, 1576–1583.
- Vile, M.A., Bridgman, S.D., Wieder, R.K., 2003a. Response of anaerobic carbon mineralization rates to sulfate amendments in a boreal peatland. *Ecol. Appl.* 13, 720–734.
- Vile, M.A., Bridgman, S.D., Wieder, R.K., Novak, M., 2003b. Atmospheric sulfur deposition alters pathways of gaseous carbon production in peatlands. *Glob. Biochem. Cycles* 17, 1058–1064.
- World Reference Base for Soil Resources, 2014. International soil classification system for naming soils and creating legends for soil maps. *World Soil Resour. Rep.* 106.
- Yavitt, J.B., Williams, C.J., Wieder, R.K., 2000. Controls on microbial production of methane and carbon dioxide in three sphagnum-dominated peatland ecosystems as revealed by a reciprocal field peat transplant study. *Geomicrobiol. J.* 39, 194–204.
- Ye, R., Jin, Q., Bohannon, B., Keller, K.J., McAllister, A.S., Bridgman, D.S., 2012. pH controls over anaerobic carbon mineralization, the efficiency of methane production, and methanogenic pathways in peatlands across an ombrotrophic-minerotrophic gradient. *Soil Biol. Biochem.* 54, 36–47.
- Yu, Z.S., 2012. Northern peatland carbon stocks and dynamics: a review. *Biogeosciences* 9, 4071–4085.